

**Economic Geology Report 5**

**BROMIDE, IODIDE, AND BORON  
IN ALBERTA FORMATION WATERS**

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## CONTENTS

	<b>Page</b>
<b>Abstract</b> .....	1
<b>Introduction</b> .....	1
Acknowledgements .....	1
<b>Analytical procedures</b> .....	1
<b>Composition of commercial brines</b> .....	3
Bromine production .....	3
Iodine production .....	3
Boron production .....	5
<b>Bromide in Alberta formation waters</b> .....	5
<b>Iodide in Alberta formation waters</b> .....	8
<b>Boron in Alberta formation waters</b> .....	13
<b>Conclusions</b> .....	14
<b>References cited</b> .....	14
<b>Appendix: Bromide, iodide and boron contents of formation water samples</b> .....	17

## FIGURES

Figure 1. Bromide in formation waters, Upper Devonian Winterburn Group, Alberta .....	6
Figure 2. Bromide in formation waters, Upper Devonian Woodbend Group, Alberta .....	6
Figure 3. Bromide in formation waters, Upper Devonian Beaverhill Lake Formation, Alberta ...	6
Figure 4. Bromide in formation waters, Middle Devonian Elk Point Group, Alberta .....	6
Figure 5. Iodide in formation waters, Lower Cretaceous lower Colorado Group, Viking and Bow Island Formations, Alberta .....	11
Figure 6. Iodide in formation waters, Upper Cretaceous post-Colorado Supergroup, Belly River Formation, Alberta .....	12
Figure 7. Iodide in formation waters, Pennsylvanian-Permian strata, Alberta .....	13
Figure 8. Scatter diagram of iodide and total dissolved solids in formation waters from Jurassic, Triassic, Pennsylvanian-Permian and Mississippian strata, northwestern Alberta ..	13

## TABLES

Table 1. Comparison of analyses of bromide in duplicate samples made by the Alberta Energy Resources Conservation Board and the University of Calgary. ....	2
Table 2. Comparison of analyses of bromide in duplicate samples made by the Dow Chemical Company and the University of Calgary .....	2
Table 3. Analyses of sea water and commercial brines .....	4
Table 4. Formation waters from Alberta wildcat wells with 1250 mg/l or more bromide. ....	7
Table 5. Formation waters from Alberta with 40 mg/l or more iodide .....	9

# BROMIDE, IODIDE, AND BORON IN ALBERTA FORMATION WATERS

## ABSTRACT

More than 860 formation waters from Alberta were analyzed for bromide, iodide, and boron. Bromide contents up to 2786 mg/l were found in high calcium and magnesium brines associated with evaporites in the Upper Devonian Beaverhill Lake Formation and Middle Devonian Elk Point Group. The most extensive regions of high-iodide formation waters are in the Cretaceous Viking, Bow Island, and basal Belly River Formations, with iodide contents generally in the range of 40 to 50 mg/l but reported up to 80 mg/l. Isolated occurrences of formation waters with iodide contents in the range of 47 to 59 mg/l were found in the Lower Jurassic Nordegg Member equivalent and Pennsylvanian-Permian strata, and are possibly geochemically and genetically associated with phosphate rocks. Boron contents up to 920 mg/l have been found in Alberta formation waters, possibly the highest concentration in any subsurface water.

## INTRODUCTION

Bromide, iodide and boron\* are present in essentially all formation waters and are occasionally found in amounts of economic importance. Iodide and, to a lesser extent, bromide have been shown to be petroleum indicators in formation waters from Alberta (Hitchon and Horn, 1974). For these reasons, bromide, iodide, and boron were determined by the authors in a suite of more than 860 formation waters from Alberta; the results are given in the appendix to this publication. This information complements that published by Hitchon and Holter (1971) on calcium and magnesium in Alberta brines, and is part of a continuing study of the economic aspects of formation waters from Alberta. Further, it supplements data used in the discriminant function study of Hitchon and Horn (1974), inasmuch as a portion of the data presented here was used in that statistical study. The chloride contents given in the appendix were determined by the Alberta Energy Resources Conservation Board, and these, together with locations and depths of the samples, may be used to locate the complete analyses in the files of the Board. It is the intention of this report to describe briefly the economically more interesting aspects of bromide, iodide, and boron in Alberta formation waters.

\*Bromine and iodine are found in formation waters effectively only as bromide and iodide ions, respectively; boron is found mainly as borates. The two halogens were analyzed in the elemental form; boron was analyzed as a tetraborate complex. For convenience in this report these components of formation waters will be referred to as bromide, iodide and boron.

## ACKNOWLEDGMENTS

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## ANALYTICAL PROCEDURES

Bromide was determined by the X-ray fluorescence method of Dunton (1968) who concluded that in the concentration range 1 to 1000 ppm this method is as accurate as the most commonly used chemical methods. A similar X-ray fluorescence procedure for the determination of bromide in waters was described by Deutsch (1974). Our working curve was prepared using synthetic standards and formation waters previously analyzed by the Alberta Energy Resources Conservation Board (Table 1) and The Dow Chemical Company (Table 2).

Iodide was determined by the sodium thiosulfate titrimetric method, as modified by D. R. Shaw, Chief Chemist, Alberta Energy Resources Conservation Board, and our results compared favorably with analyzed formation waters supplied by The Dow Chemical Company and the Alberta Energy Resources Conservation Board.

**Table 1. Comparison of analyses of bromide in duplicate samples made by the Alberta Energy Resources Conservation Board and the University of Calgary**

Sample Number	The Alberta Energy Resources Conservation Board	The University of Calgary (A. A. Levinson)
W5	219	212
W13	296	318
W17	466	508
W20	193	215
W26	20	8
W28	35	5
W30	375	392
W32	820	950
W38	996	955
W45	961	917
W50	1121	1080
W52	13	0
W58	74	85
W66	89	103
W68	81	95
W70	trace	0
W79	459	420
W81	76	100
W85	2	0
W88	trace	0
W90	516	470
W92	267	248

**Table 2. Comparison of analyses of bromide in duplicate samples made by The Dow Chemical Company and the University of Calgary**

Sample Number	The Dow Chemical Company	The University of Calgary (A. A. Levinson)
DC 633	186	215
DC 628	1112	1155
DC 622	2251	2225

Boron was determined spectrophotometrically using the methylene blue — boron tetrafluoride method which was modified from that described by Pasztor, Bode, and Fernando (1960) and Schleicher and Heck (1970). Specifically, the procedure is:

From 1 to 10 ml of formation water (which has been diluted 1/10 to 1/20 of the original formation water) is pipetted into a 100 ml polyethylene bottle. Add 10 ml of 2.5 N H<sub>2</sub>SO<sub>4</sub> and dilute with distilled water to 20 ml. Add 5 ml of 5 percent HF, and let stand for at least 2 hours. Dilute to 50 ml and add 10 ml of 0.001 M methylene blue. Add 25 ml ethylene dichloride, stopper, and shake vigorously for at least 1 minute. Let the organic layer separate, and from the organic layer pipet 5 ml and transfer to a 25 ml volumetric flask which is made up to 25 ml with ethylene dichloride. Measure the absorbance at 660 m $\mu$  (we used a Beckman DU-2 spectrophotometer). A blank was run with each series of analyses. Chlorine, when present in large amounts, has been reported to interfere with this colorimetric procedure; specifically, chlorine will result in apparently higher values for boron. Several experiments were carried out in which we found that because it was necessary to dilute the original formation water to keep the color complex formation within the range of the standard working curve, chlorine did not interfere to any measurable degree.

### COMPOSITION OF COMMERCIAL BRINES

Sea water, formation waters, and brines from saline lakes are exploited commercially for a variety of components, including bromine, iodine and boron, at several places throughout the world. Analyses of sea water and some of these commercial brines are shown in table 3.

### BROMINE PRODUCTION

All methods of bromine production depend on the oxidation of the bromide ion which is found naturally only in relatively low concentrations in sea water (67 mg/l), formation waters, and brines from saline lakes. In the commercial isolation of bromine from these natural sources, there are four essential steps:

- (1) oxidation of bromide to bromine
- (2) removal of bromine vapor from the solution
- (3) condensation of the vapor or fixation in some chemical form
- (4) purification of the product.

Chlorine is the only oxidizing agent employed commercially in step (1) although other agents and techniques have been used in the past. Step (2) involves driving out the bromine vapor with a current of air or steam. Steam is suitable when the raw brine is relatively rich in bromine (1000 mg/l or more), but air is more economical when the bromine is extracted from very dilute solutions, such as sea water. Simple condensation of the vapor yields crude bromine which is subsequently purified. If alkali or alkaline earth bromides or bromates are to be manufactured, step (3) involves reaction

of the vapor with the appropriate metal carbonate or hydroxide, followed by suitable purification of the metal salt. Further details of these and other processes related to the recovery of bromine may be found in Downs and Adams (1973).

The United States produces and consumes about three fourths of the world bromine supply, and of this production over three fourths comes from Jurassic Smackover Formation brines in Arkansas, which contain over 5000 mg/l bromide (Table 3). Until this relatively very recent production from Arkansas, effectively the only source of bromine production in the United States was from Midland, Michigan, where the brines contain about 2900 mg/l bromide (Table 3), with only very minor production from the Searles Lake brines as a byproduct of potassium chloride production (Ryan, 1951). Next to the United Kingdom, Israel is now the world's third largest producer of bromine with virtually unlimited resources in the Dead Sea, estimated by Neev and Emery (1967) at  $0.694 \times 10^9$  metric tons. According to Kenat (1966) the brines in the solar evaporation ponds contain about 12,000 mg/l bromide when they are drawn off for bromine production during the carnallite crystallization stage. The United Kingdom, Japan, and Italy continue to produce bromine from sea water, but in France, bromine is recovered as a coproduct at the potash mines in Mulhouse in the Alsace area, although apparently production from these mines is limited by law to prevent excessive damage to the environment (Klingman, 1974). The USSR produces bromine at Nebit-Dag in Turkmenia from formation waters and from the Kara Bogaz. There is no Canadian production of bromine. If the plans reported in financial papers materialize, then Israel would be second only to the United States in world bromine production by the end of this decade, accounting for one sixth of the world production. The significance of this, and the recent pre-eminence of Arkansas as a source of bromine in the United States, to potential bromine production in Alberta suggests that formation waters should contain at least 3000 mg/l and preferably 5000 mg/l bromide before they can be considered a potential economic source of bromide, all other factors being equal.

### IODINE PRODUCTION

Iodine may be recovered by one of several possible processes, including a process resembling that for the recovery of bromine from sea water in which the brine is acidified with sulfuric acid and treated with a slight excess of chlorine to liberate iodine. The liquor is subsequently stripped of its iodine by a current of air which then passes to a tower where the iodine is absorbed by a solution of hydriodic and sulfuric acids into which sulfur dioxide is passed continuously to reduce the iodine to iodide. When sufficiently concentrated, this solution is again chlorinated; the iodine precipitates and is then filtered off. Two alternate recovery processes involve the precipitation of insoluble metal iodides. In one process,

Table 3. Analyses of sea water and commercial brines

Sample Details	Chemical Constituents (mg/l)										Total Dissolved Solids	Reference
	Na	K	Ca	Mg	Cl	Br	I	B	SO <sub>4</sub>	HCO <sub>3</sub>		
Sea water (standard salinity of 35‰)	10,760	387	413	1,294	19,353	67	0.06	4	2,712	142	35,000	Culkin, 1965 (Table 1)
Midland, Michigan (Lower Devonian Sylvania Sandstone)	22,500	9,120	74,800	9,960	208,000	2,910	40	380	40	**	331,000 <sup>1</sup>	White, Hem, and Waring, 1963 (Table 13, No. 8)
Columbia County, Arkansas (Jurassic Smackover Formation)	70,850	6,380	44,851	3,841	210,715	5,249	10	288	174	191	346,139 <sup>2</sup>	A. G. Collins (pers. comm.)
Searles Lake, California	111,000	23,900	*	*	117,000	704	27	3,420	37,800	**	330,000 <sup>3</sup>	Y. K. Kharaka (pers. comm.)
Dead Sea, Israel	39,150	7,260	16,860	40,650	212,400	5,120	< 1	~ 20	470	220	322,130	Neev and Emery, 1967 (Tables 6 and 9)
Japan	9,800	337	141	399	16,450	96	95	13	0	920	28,446 <sup>4</sup>	Marsden and Kawai, 1965 (Table 11)
Woodward County, Oklahoma (Pennsylvanian Morrow Formation)	7,210	37	649	5	16,563	438	696	31	938	334	27,313 <sup>5</sup>	A. G. Collins (pers. comm.)

\* below detection limits

\*\* no data available

1 includes Sr 2,650; NH<sub>4</sub> 506; Li 70; Fe 22; Mn 2.2 includes Sr 2,470; NH<sub>4</sub> 170; Li 391; Ba 20; Mn 46; Fe 10; Zn 2; Cu 1.3 includes CO<sub>3</sub> 32,300; SiO<sub>2</sub> 85; As 102; F 23; PO<sub>4</sub> 536.4 includes NH<sub>4</sub> 195.5 includes NH<sub>4</sub> 36; Sr 27; Li 1.

a silver nitrate solution is added in sufficient quantity to precipitate only silver iodide, which is filtered off and treated with clean steel scrap to form metallic silver and a solution of ferrous iodide. The former is processed for recycling and the ferrous iodide solution is treated with chlorine to liberate the iodine. In another process, the brine is treated with chlorine gas to liberate free iodine and is then passed over bales of copper wire with which the iodine reacts to form insoluble cuprous iodide, which is filtered, dried, and stored or transported as such. Recently developed ion-exchange techniques have been reported to give a new, more economical extraction process for producing iodine of exceptionally high purity. The process used by the Chilean nitrate industry differs from all other processes because the iodine occurs as an iodate in caliche. Downs and Adams (1973) provide more details of these and other processes related to the recovery of iodine.

Commercial iodine sources include caliche deposits in Chile (for many years the world's major source of iodine), formation waters associated with shallow gas deposits, and deep saline formation waters. The very low content of iodine in sea water has so far generally precluded its use as a commercial source, although there is an unconfirmed report (Wang, 1974) that iodine has been produced at the Haifang seawater salt processing plant in Amoy, Fukieh Province, People's Republic of China. Prior to 1964, The Dow Chemical Company extracted iodine from California oilfield brines with iodine contents reported at 60 to 70 mg/l (Anonymous, 1964). Since 1964, The Dow Chemical Company has recovered iodine from formation brines at Midland, Michigan, using improved technology so that the process is economic at an iodide content of 40 mg/l (Table 3). The Midland brines also offer an advantage in that they are free of oil, and the crude iodine is recovered as a coproduct with bromine, calcium and magnesium compounds, and potash. This operation will shortly cease to be the only iodine producer in the United States. Although the United States continues to import much of its iodine supply, formerly mainly from Chile but more recently almost solely from Japan, plans call for an iodine extraction plant at Woodward, Oklahoma, which will recover iodine from formation waters at a rate sufficient to supply about one quarter of the United States' consumption within one year; other plants in Oklahoma may follow. No analyses of formation water feedstocks from Oklahoma have been published but Collins and Egleson (1967) and Collins (1969) have reported iodide contents up to 1400 mg/l in comparable Oklahoma formation waters. Collins, Bennett, and Manuel (1971) have demonstrated an association between these high-iodide brines and the presence of algae in adjacent sedimentary rocks. There is no Canadian production of iodine.

Japan has been the world's primary producer of iodine since the late 1960's, with production coming from brackish formation waters associated with shallow gas deposits and with iodide contents reported in the range 80 to 100 mg/l. Many

of the gas fields are of the "suiyōsei-ten'nengasu" type, which translates as "natural gas which is dissolved in formation water" (Marsden and Kawai, 1965). An analysis of the brackish formation water from the important Mobarra gas field is given in table 3. The close genetic relation between iodine and natural gas has been observed by several authors who have studied these Japanese occurrences, and Motojima (1963, 1964) has evaluated the various geochemical criteria for prospecting for these gas fields, and for their related iodine resources (Motojima, 1971). Plans have been made for the development, by joint Japanese-Indonesian interests, of comparable brackish waters in Indonesia with iodide contents of more than 70 mg/l. In the USSR, iodine apparently is produced at the Neftechlinski field near the Black Sea and at a plant in the Baku area (Wang, 1974), as well as at Kopet-Dag in Turkmenia, where Kudel'skii (1970) has reported iodide contents of up to 462 mg/l. With respect to potential iodide resources in Alberta, it seems likely that formation waters with iodide contents greater than 40 mg/l may be of interest.

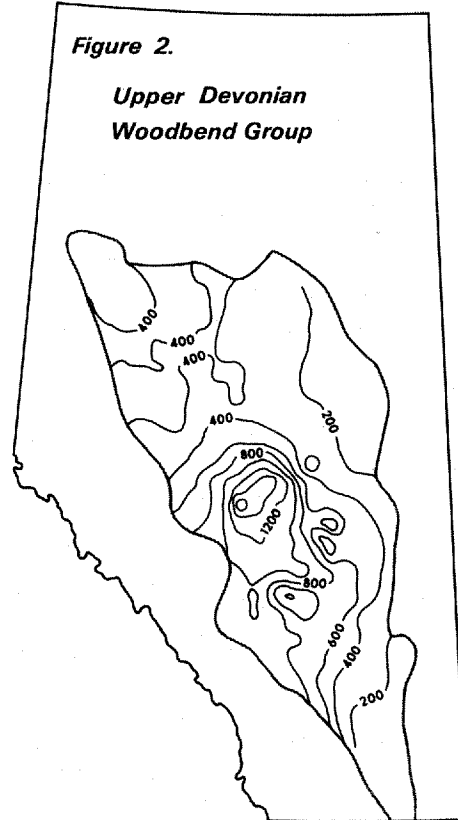
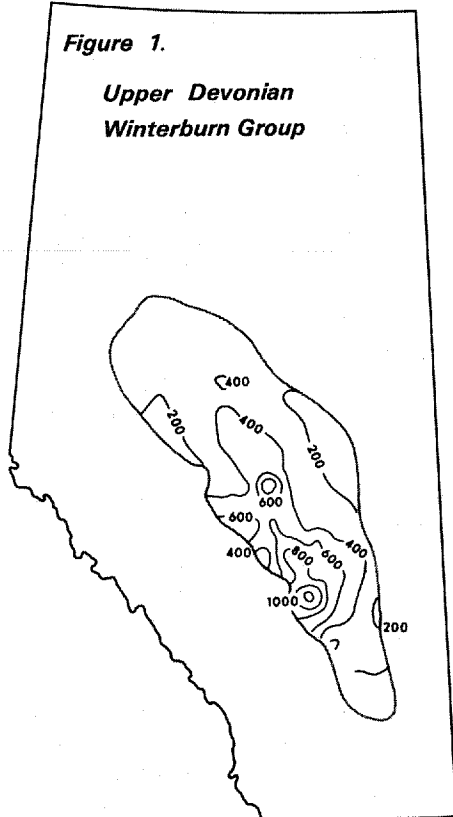
#### BORON PRODUCTION

Unlike bromide and iodide, which are effectively only recovered from aqueous solutions, boron is principally mined from deposits of borax ( $\text{Na}_2\text{B}_4\text{O}_7 \cdot 10\text{H}_2\text{O}$ ), kernite ( $\text{Na}_2\text{B}_4\text{O}_7 \cdot 4\text{H}_2\text{O}$ ), or colemanite ( $\text{Ca}_2\text{B}_6\text{O}_{11} \cdot 5\text{H}_2\text{O}$ ). However, at Searles Lake, San Bernardino County, California, a brine in equilibrium with shallow deposits of trona ( $\text{Na}_3\text{H}(\text{CO}_3)_2 \cdot 2\text{H}_2\text{O}$ ) is extracted for processing of the dissolved salts, including borates. Evaporation of the brine results in the precipitation of sodium chloride and burkeite ( $\text{Na}_6(\text{CO}_3)(\text{SO}_4)_2$ ), whereby the liquor becomes increasingly concentrated with respect to potassium chloride and sodium tetraborate. Following crystallization of potassium chloride by cooling to 100° F, and its subsequent removal through settling and filtration, the liquor is further cooled to about 80° F at which temperature crude borax is precipitated and thickened. The crude borax filter cake is brine-leached to remove soluble phosphate impurities and then recrystallized and refined. Ryan (1951) provides more details of the entire operation at Searles Lake. Because of the abundance of mineable borax deposits and the fact that boron is only recovered as a by-product of processing the Searles Lake brine, it seems highly unlikely that any formation water with less than several thousand mg/l boron could be considered a potential source of this element.

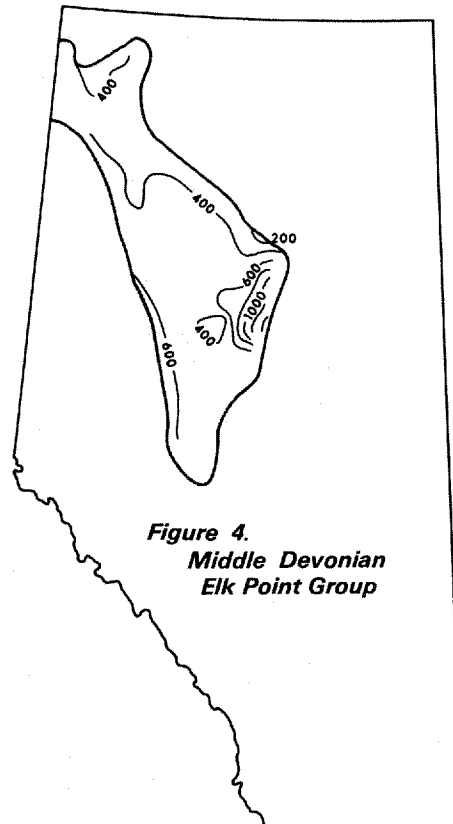
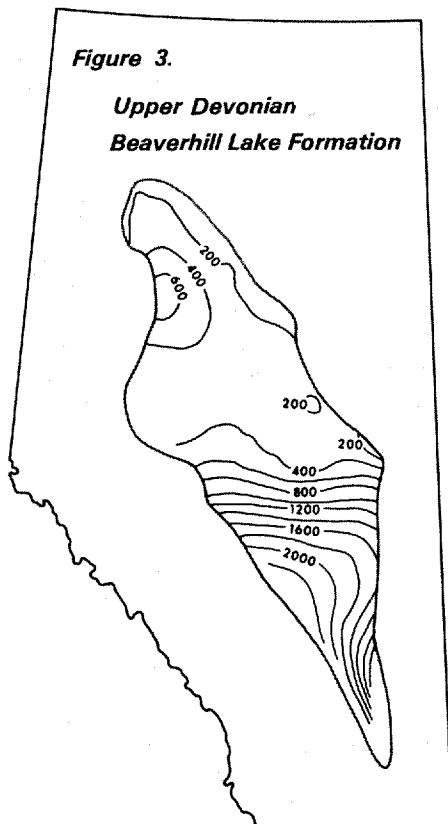
#### BROMIDE IN ALBERTA FORMATION WATERS

Bromide contents greater than 1000 mg/l are found only in formation waters from the Upper Devonian Winterburn Group, Woodbend Group, and Beaverhill Lake Formation; the Middle Devonian Elk Point Group; and the Granite Wash. The computer-contoured maps in figures 1 to 4 show the distribution of bromide in formation waters from the four





*Bromide (mg/l) in formation waters, Alberta*



cited Devonian stratigraphic units, respectively. The computer data bank includes information from Shaw (1962), Hitchon, Billings and Klován (1971), Hitchon and Holter (1971), Hitchon and Horn (1974), the files of the Alberta Energy Resources Conservation Board, the appendix of this report, and unpublished analyses in the files of the senior author. All formation waters from producing oil and gas fields have bromide contents less than 1500 mg/l. The highest recorded amount is 2786 mg/l from a wildcat well in the Beaverhill Lake Formation of southern Alberta. These observations suggest that formation waters with high bromide contents from so-called dry holes may have the greatest potential for the production of bromide. They have the further economic and technical advantage of not requiring the clean-up of the coproduced formation water from entrained crude oil. Table 4 provides details of formation waters with 1250 mg/l or more bromide from wildcat wells in Alberta. The information in the maps, table 4, and the computer data bank, indicates three areas of interest. The most important is the Beaverhill Lake Formation of southern Alberta with formation waters from four locations having bromide contents greater than 1500 mg/l. This is approximately the same region which has formation waters with high contents of calcium and magnesium (Hitchon and Holter, 1971). Absence of iodide from some of these high-bromide samples suggests

a genetic geochemical association with the evaporites which are present within the Beaverhill Lake Formation of southern Alberta and in nearby underlying Middle Devonian Elk Point Group rocks. A second area of interest comprises a small, isolated portion of the Cooking Lake Formation carbonate platform in central Alberta, south of Edmonton and east of the Homeglen-Rimbey reef trend of the Leduc Formation, where the formation waters contain up to 1650 mg/l bromide. The third region of interest is the Keg River Formation of north central Alberta which is only sparsely drilled and for which only one high bromide formation water is available. However, the association of high calcium and magnesium brines (Hitchon and Holter, 1971) and evaporites in the area make it worthy of exploration for high bromide formation waters.

From the previous comments on commercial brines, it is clear that no Alberta formation water contains as much bromide as the Midland, Michigan brines, and all are far below the 5000 mg/l of the now dominant sources in Arkansas and the Dead Sea. On this basis alone it seems highly unlikely that any of the Alberta formation waters discovered so far would be an economic source of bromide. This conclusion should not preclude analyzing all formation waters for bromide and iodide, especially those recovered from areas that may contain potentially valuable brines.

Table 4. Formation waters from Alberta wildcat wells with 1250 mg/l or more bromide

Lsd	Location				Depth (ft)	Sampling Conditions	Stratigraphic Unit	Br (mg/l)	I (mg/l)
	Sec	Tp	R	Mer					
10	21	13	11	W4	5205-5225	Drillstem Test: 350 ft sw	Beaverhill Lake Formation	2786	0
4	17	30	20	W4	6960-7070	Drillstem Test: 860 ft sw	Beaverhill Lake Formation	2115	27
4	34	37	26	W4	7405-7462	Drillstem Test: 2500 ft sw	Leduc Formation	2115	2
11	7	36	23	W4	7650-7700	Drillstem Test: 4450 ft sw	Beaverhill Lake Formation	1900	0
13	16	50	24	W4	5620-5650	Drillstem Test: 670 ft sw	Cooking Lake Formation	1650	31
15	16	44	13	W4	4272-4316	Drillstem Test: 125 ft sw, 122 ft mud	Beaverhill Lake Formation	1590	25
4	3	88	25	W4	4278-4316	Drillstem Test: 3620 ft sw	Keg River Formation	1475	18
7	30	50	22	W4	5408-5431	Drillstem Test: 1260 ft gsw	Cooking Lake Formation	1470	34
12	14	12	12	W4	5120-5202	Drillstem Test: 518 ft sw	Beaverhill Lake Formation	1375	0
13	30	48	22	W4	5610-5650	Drillstem Test: 60 ft sw, 150 ft mud	Cooking Lake Formation	1250	21

## IODIDE IN ALBERTA FORMATION WATERS

There is an ill-defined stratigraphic dichotomy between those formation waters with economically interesting bromide contents and other formation waters with potentially productive iodide concentrations. As noted previously, the former are confined to Devonian and Granite Wash strata. If we arbitrarily define the latter as formation waters with iodide contents 40 mg/l, the same concentration as the Michigan brines (Table 3), then formation waters with potentially productive iodide concentrations are effectively found only in Lower Cretaceous lower Colorado Group (Viking and Bow Island Formations) and Upper Cretaceous post-Colorado Supergroup (basal Belly River Formation) strata. Isolated examples of formation waters with more than 40 mg/l iodide are recorded from four stratigraphic units — Upper Cretaceous Medicine Hat Sandstone, Lower Jurassic Nordegg Member equivalent, Pennsylvanian-Permian strata, and Upper Devonian Slave Point Formation. Table 5 provides details of formation waters with 40 mg/l or more iodide and has been compiled using data from Shaw (1962), Hitchon and Horn (1974), the files of the Alberta Energy Resources Conservation Board, the appendix of this report, and unpublished analyses in the files of the senior author.

The map in figure 5 shows the distribution of iodide in formation waters from the Viking and Bow Island Formations. In four regions the formation waters contain more than 40 mg/l iodide, although subsequent data may show the two northern regions to be contiguous. Contents range up to 80 mg/l, though most are in the order of 40 to 50 mg/l. Many of the formation water samples are from drillstem tests, with all the uncertainties to which that method of sampling is prone, but nevertheless a sufficient number are from producing wells, so there can be little doubt about the broad regional pattern. The northwest orientation of some of the potentially productive iodide regions is probably related to the depositional environment of the Viking and Bow Island Formations as reflected in the total sandstone isoliths (Rudkin, 1964), an observation which may be of interest in subsequent exploration for economic iodide concentrations.

Iodide contents up to 54 mg/l have been found in formation waters from sandstones near the base of the Belly River Formation, with most samples from the Pembina field. Figure 6 shows the regional variations in the iodide concentration of formation waters from this stratigraphic unit. In addition, formation waters in the basal sandstones are distinguished from those elsewhere in the formation because of an incompletely confirmed tendency for higher iodide contents to occur in the formation waters from the basal sandstones. Paucity of data precludes the determination of a relation between high-iodide trends and lithology, such as was tentatively cited for the Viking and Bow Island Formations.

Isolated examples of formation waters with more than 40 mg/l iodide have been discovered in the Permian Belloy Formation and the subjacent Pennsylvanian Kiskatinaw Formation, in adjacent townships close to the border between Alberta and British Columbia. Details of the samples are provided in table 5 and their location with respect to the regional variations in iodide content of formation waters from Pennsylvanian-Permian strata is shown in figure 7. Study of the relations between the iodide content of formation waters from Pennsylvanian-Permian strata and those of the overlying Triassic and underlying Mississippian rocks reveals generally higher contents of iodide in the former compared to the latter, for any given salinity. These relations are illustrated in figure 8. Goldschmidt (1954) has observed that marine phosphate sediments contain surprisingly large amounts of iodine, and thus the presence of phosphates in Permian rocks and their absence from Triassic and Mississippian strata may explain the relations shown in figure 8. In this regard, it is interesting to observe that the single sample of formation water with a high iodide content from the Lower Jurassic is from the Nordegg Member equivalent. According to Springer, MacDonald, and Crockford (1964), the lithology of Sinemurian rocks, of which the Nordegg Member forms a distinct facies, changes from region to region and consists of phosphatic shale, coquina and conglomerate, or an impure, cherty, phosphatic limestone known as the Nordegg Member. The iodine-rich single sample is from thick shales equivalent to the Nordegg Member. Any search for high-iodide formation waters in Lower Jurassic or Pennsylvanian-Permian rocks should be cognizant of the geochemical relation between iodine and phosphate.

Two other stratigraphic units have yielded formation waters with iodide contents of 40 mg/l or more. They are an isolated sample from the Medicine Hat Sandstone in the Medicine Hat field and two samples from the Slave Point and Beaverhill Lake Formations in the Swan Hills reef trend, which have iodide contents of 41 and 40 mg/l respectively, in a region in which the iodide content of adjacent oil field and gas field waters is in the range 18 to 39 mg/l.

To summarize the occurrence of economically interesting high-iodide formation waters in Alberta: the most extensive regions are in the Viking, Bow Island, and basal Belly River Formations with iodide contents generally in the range 40 to 50 mg/l. Isolated occurrences of formation waters with iodide contents in the range 47 to 59 mg/l are found in the Lower Jurassic Nordegg Member equivalent and Pennsylvanian-Permian strata, and are possibly geochemically and genetically associated with the presence of phosphate rocks. The only other potentially interesting sample is an isolated example from the Medicine Hat Sandstone. The obviously wide diversity of occurrence of iodide-rich formation waters and paucity of data in certain regions suggests that a special exploratory effort should be made to determine the source and regional limits of these iodide occurrences in Alberta.

Table 5. Formation waters from Alberta with 40 mg/l or more iodide

Lsd	Location			Mer	Depth (ft)	Sampling Conditions	Stratigraphic Unit	I (mg/l)	Br (mg/l)
	Sec	Tp	R						
10	33	10	18	W4	2517-2542	Drillstem Test: 1910 ft sw, 60 ft mud	Bow Island Formation	80	122
11	14	12	2	W4	1516-1536	Gas well	Medicine Hat Sandstone	62	87
10	21	74	18	W5	3164-3180	Drillstem Test: 435 ft sw, 100 ft mud	Nordeg Member Equivalent	59	14
-	-	48	3/4	W5	3206-3326	Oilfield battery	Basal Belly River Formation	54	139
6	19	40	16	W4	1490-1576	Drillstem Test: 1076 ft sw	Basal Belly River Formation	53	105
8	2	11	17	W4	2210-2314	Drillstem Test: 1600 ft sw	Bow Island Formation	53	88
7	20	23	18	W4	3404-3423	Gas well	Viking Formation	51	117
11	5	40	19	W4	3873-3933	Drillstem Test: 900 ft sw, 20 ft mud	Viking Formation	49	235
4	7	42	22	W4	4197-4244	Drillstem Test: 1680 ft sw	Viking Formation	49	206
10	14	29	20	W4	3549-3595	Drillstem Test: 1960 ft sw	Viking Formation	49	180
14	35	47	3	W5	3309-3317	Oil well	Basal Belly River Formation	49	143
11	10	80	12	W6	7914-7935	Drillstem Test: 874 ft sw, 92 ft mud	Kiskatinaw Formation	47	166
10	13	81	13	W6	6500-6590	Drillstem Test: 4700 ft sw	Belloy Formation	47	160
6	5	27	7	W4	3436-3448	Drillstem Test: 2665 ft sw	Viking Formation	47	113
10	23	35	12	W4	3035-3135	Drillstem Test: 1620 ft sw, 90 ft mud	Viking Formation	47	81
1	12	52	27	W4	3590-3605	Drillstem Test: 2295 ft sw, 30 ft mud	Viking Formation	46	310
7	18	41	23	W4	4401-4419	Drillstem Test: 180 ft oc sw, 100 ft mud	Viking Formation	46	230
10	8	20	14	W4	2837-2870	Drillstem Test: 1750 ft sw	Bow Island Formation	46	205
11	4	23	14	W4	2553-2600	Drillstem Test: 210 ft sw, 60 ft mud	Bow Island Formation	46	170
14	5	49	22	W4	3425-3445	Drillstem Test: 1765 ft sw	Viking Formation	45	290

Table 5. (continued)

Lsd	Location				Mer	Depth (ft)	Sampling Conditions	Stratigraphic Unit	I (mg/l)	Br (mg/l)
	Sec	Tp	R							
7	34	41	19	W4	W4	3500-3550	Drillstem Test: 900 ft sw, 80 ft mud	Viking Formation	45	260
11	3	20	16	W4	W4	3032-3040	Drillstem Test: 420 ft sw, 60 ft mud	Bow Island Formation	45	203
10	12	25	23	W4	W4	4201-4226	Drillstem Test: 270 ft sl m sw, 180 ft mud	Viking Formation	45	150
6/3	30	48	6	W5	W5	3608-3742	Oilfield battery	Basal Belly River Formation	44	160
6	9	48	2	W5	W5	2990-3021	Drillstem Test: 600 ft sl m sw	Basal Belly River Formation	44	100
-	7/18	35	9	W4	W4	2900-3000	Oilfield battery	Viking Formation	44	90
13	16	50	24	W4	W4	3350-3385	Drillstem Test: 1375 ft sw, 186 ft gc sw, 465 ft gc mud	Viking Formation	43	350
4	25	48	5	W5	W5	3170-3176	Oil well	Basal Belly River Formation	43	95
10	11	46	25	W4	W4	4083-4163	Drillstem Test: 2300 ft sw, 300 ft mud	Viking Formation	42	179
14	21	48	2	W5	W5	2908-2925	Drillstem Test: 180 ft sw, 230 ft o mud	Basal Belly River Formation	42	105
4	2	63	10	W5	W5	8119-8205	Drillstem Test: 2100 ft gc sw, 110 ft oc sw, 300 ft frothy oc mud	Slave Point Formation	41	365
7	18	43	20	W4	W4	3525-3536	Drillstem Test	Viking Formation	41	224
7	14	27	21	W4	W4	3920-3950	Drillstem Test: 780 ft sw	Viking Formation	41	180
10	27	19	16	W4	W4	2800-2815	Drillstem Test: 300 ft mud	Bow Island Formation	41	160
-	-	47	4	W5	W5	-	Oilfield battery	Basal Belly River Formation	41	106
4	30	65	13	W5	W5	9372-9379	Oil well	Beaverhill Lake Formation	40	291
16	14	48	3	W5	W5	3046-3075	Oil well	Basal Belly River Formation	40	104
-	7/18	48	2	W5	W5	-	Oilfield battery	Basal Belly River Formation	40	89
6	30	10	17	W4	W4	2384-2422	Drillstem Test: 535 ft mc sw, 100 ft mud	Bow Island Formation	40	80
-	-	56	25	W4	W4	avg. 2822	Oilfield battery	Viking Formation	40	60

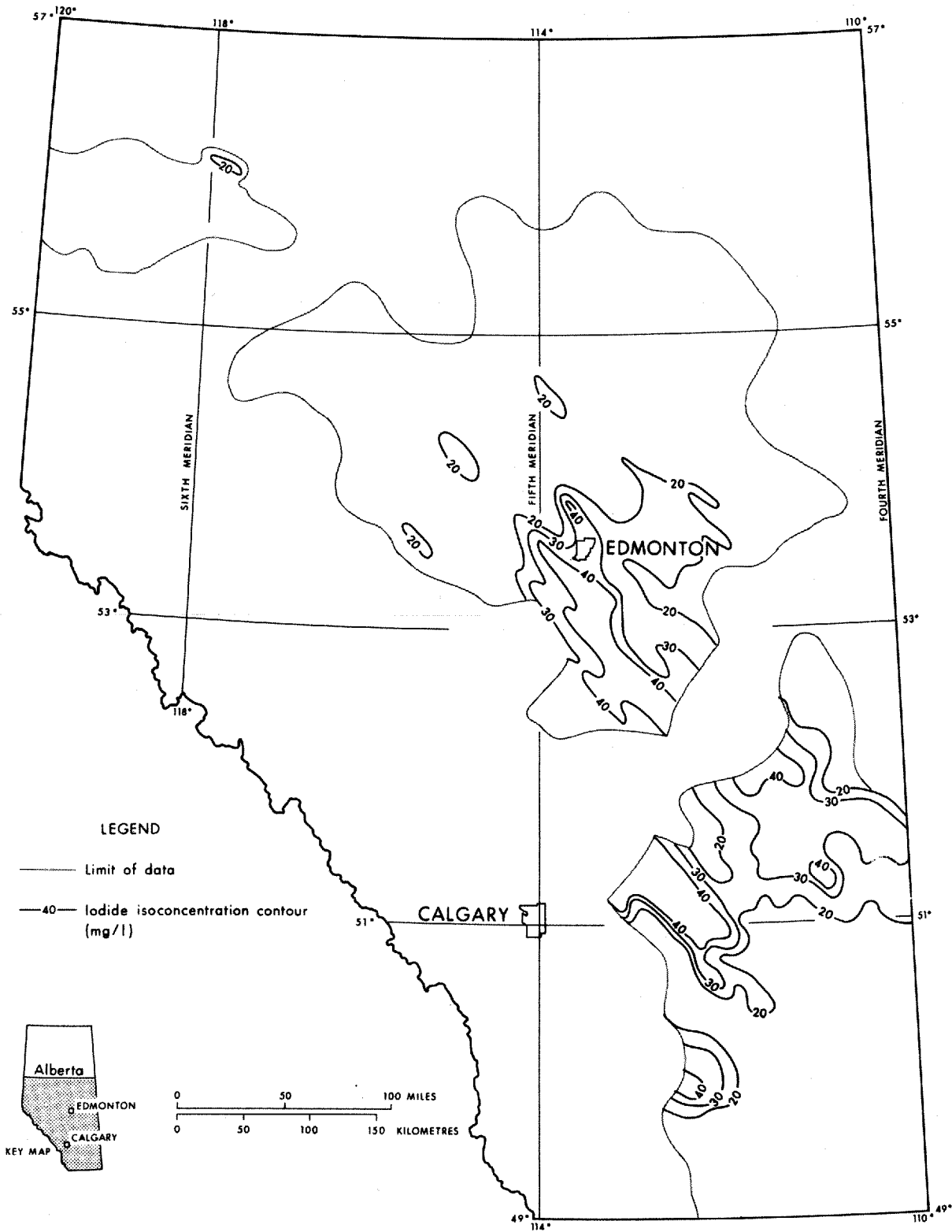


Figure 5. Iodide (mg/l) in formation waters, Lower Cretaceous lower Colorado Group, Viking and Bow Island Formations, Alberta

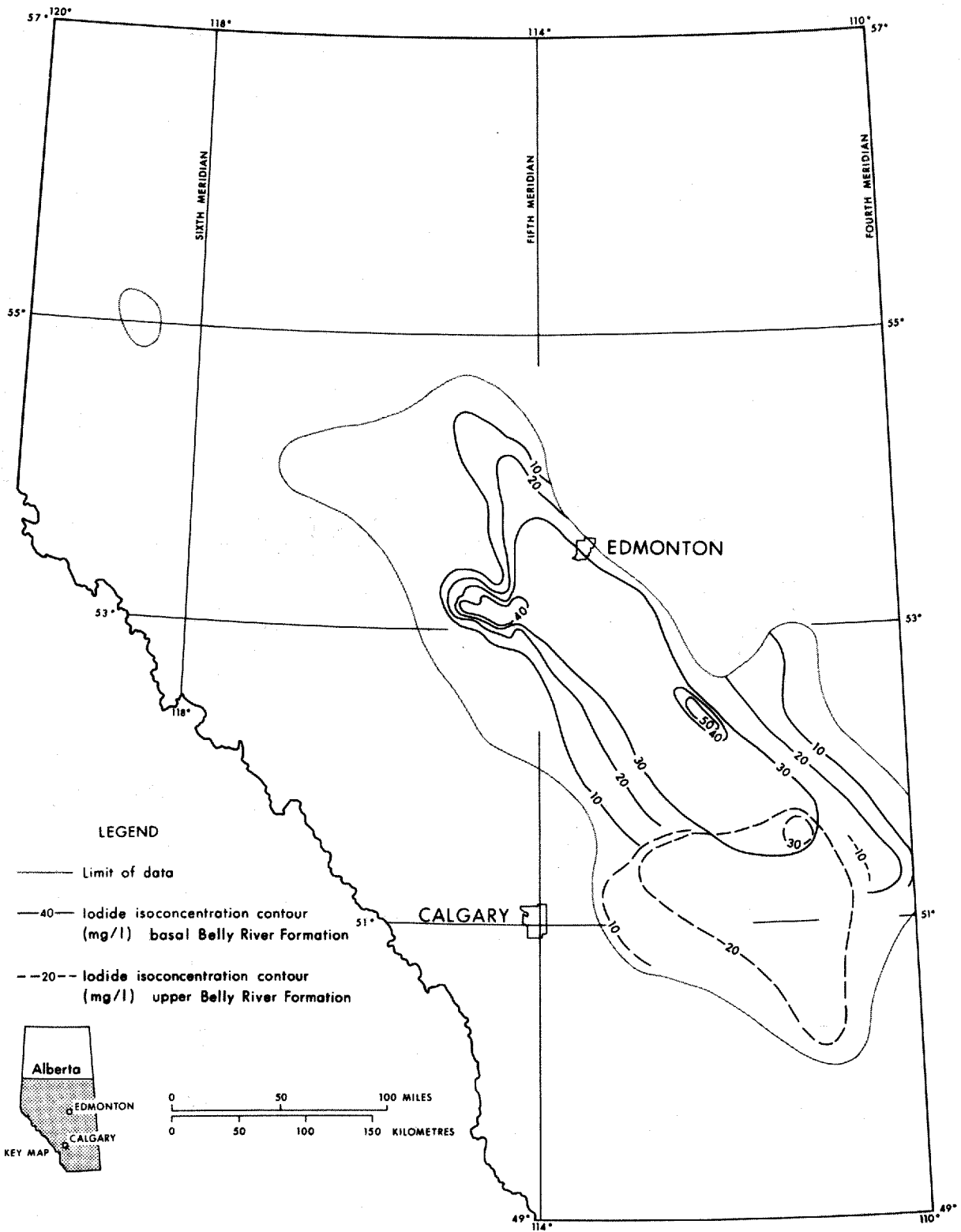
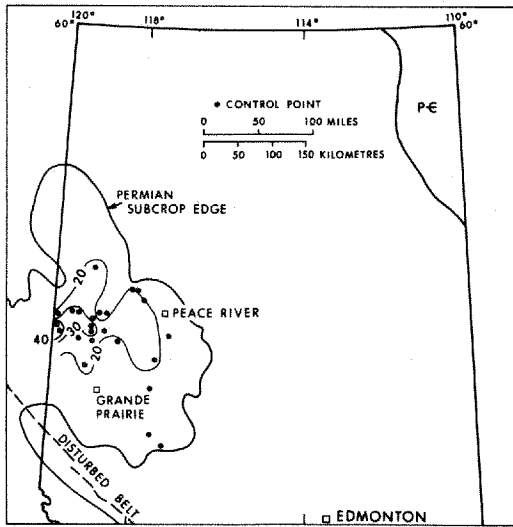


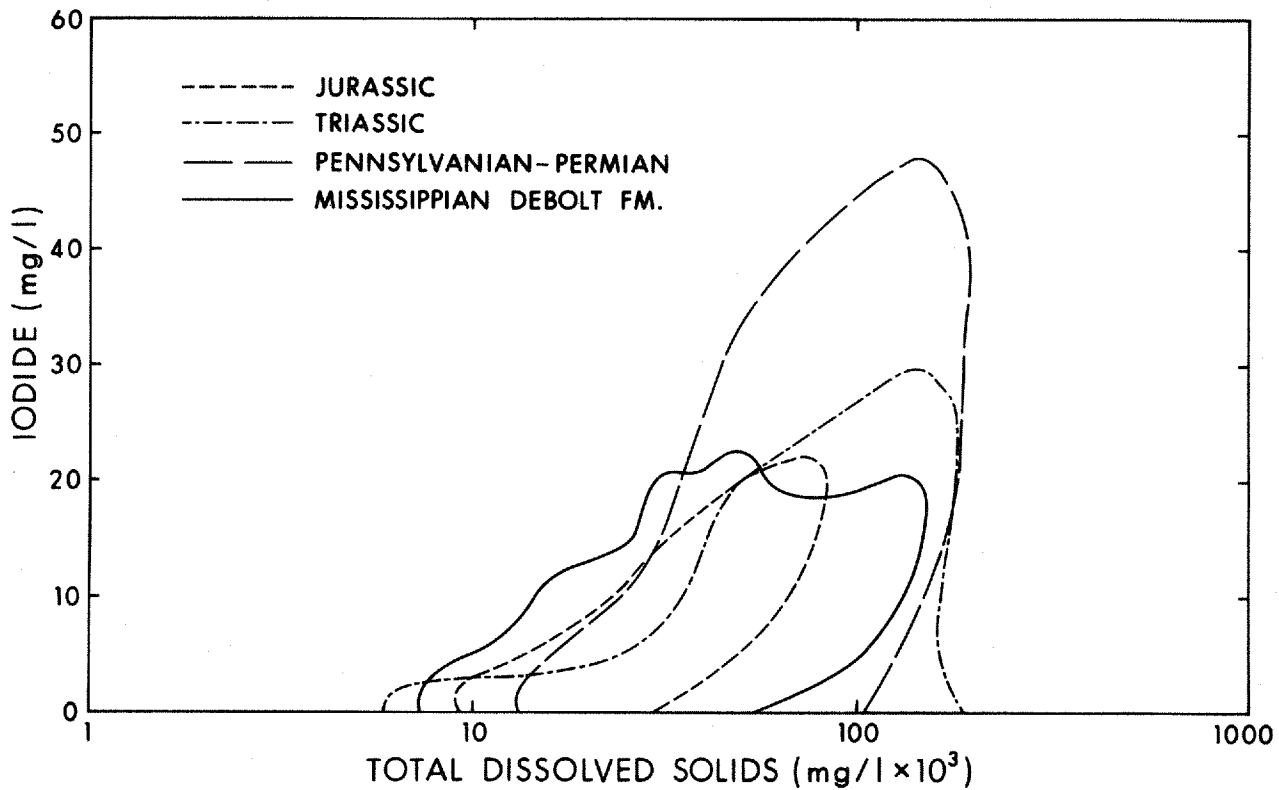
Figure 6. Iodide (mg/l) in formation waters, Upper Cretaceous post-Colorado Supergroup, Belly River Formation, Alberta

**BORON IN ALBERTA FORMATION WATERS**

The only published data on boron in Alberta formation waters are Hitchon, Billings and Klovan (1971) and the appendix of this report. Contents of boron range up to 86 mg/l and average 15.7 mg/l in formation waters from Mesozoic strata. In formation waters from Paleozoic strata the average content of boron is 101 mg/l with one sample from the Upper Devonian Beaverhill Lake Formation of north central Alberta containing 920 mg/l. A thorough, but by no means exhaustive, survey of the literature shows this latter value to be the maximum boron reported in a formation water (Hitchon, in press). All these boron concentrations are definitely not commercial at this time, nor is it very likely that commercial concentrations will be found. Nevertheless, the boron values cited in the appendix of this report may be of interest to those seeking to use boron as an indirect petroleum indicator, as suggested by several authors including Sivan (1972) and Gutsalo (1974). However, it is our opinion that until a statistical study is available on the importance of boron as a petroleum indicator, such as was reported for Alberta by Hitchon and Horn (1974), the merits of using boron as a petroleum indicator must remain equivocal.



**Figure 7. Iodide (mg/l) in formation waters, Pennsylvanian-Permian strata, Alberta**



**Figure 8. Scatter diagram of iodide and total dissolved solids in formation waters from Jurassic, Triassic, Pennsylvanian-Permian and Mississippian strata, northwestern Alberta**



### CONCLUSIONS

(1) Bromide contents up to 2786 mg/l have been reported in high calcium and magnesium brines of Alberta associated with evaporites in the Upper Devonian Beaverhill Lake Formation and the Middle Devonian Elk Point Group. This concentration is close to that at Midland, Michigan, but below the 5000 mg/l for the now dominant commercial sources in Arkansas and the Dead Sea. It seems highly unlikely that Alberta formation waters will be discovered which would be a separate economic source of bromide, although the recovery of bromide together with, say, calcium and magnesium, might prove economically viable.

(2) Iodide contents up to 80 mg/l have been reported in Alberta formation waters. The most extensive regions of high-iodide formation waters are in the Cretaceous Viking,

Bow Island, and basal Belly River Formations with iodide contents generally in the range 40 to 50 mg/l. Isolated occurrences of formation waters with iodide contents in the range 47 to 59 mg/l are found in the Lower Jurassic Nordegg Member equivalent and Pennsylvanian-Permian strata, and are possibly geochemically and genetically associated with the presence of phosphate rocks. These concentrations are greater than that at Midland, Michigan, but less than Japanese and Oklahoma iodide resources. It seems likely that Alberta formation waters with iodide contents greater than 40 mg/l may be of economic interest.

(3) Boron contents up to 920 mg/l have been reported in Alberta formation waters but no commercial exploitation is likely at this time in view of other major world sources of boron.

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**Appendix**

**BROMIDE, IODIDE AND BORON CONTENTS OF  
FORMATION WATER SAMPLES**

Upper Cretaceous, Post-Colorado Supergroup							Lower Cretaceous, Lower Colorado Group (cont.)												
Location			Interval	Cl	Br	I	B	Location			Interval	Cl	Br	I	B				
10	12	10	11	W.4	786- 830	550	0	3	10	11	11	14	W.4	2120- 2325	3112	15	6	5	
8	10	11	13	W.4	600- 700	1199	0	1	7	10	18	12	10	W.4	2105- 2175	3750	30	13	
6	25	13	17	W.4	940- 941	4188	70	32	4	6	25	13	17	W.4	2520- 2521	9440	110	36	
7	15	22	7	W.4	508- 518	2765	35	21	3	4	30	14	13	W.4	2452- 2473	4592	45	7	
7	15	22	7	W.4	600- 635	3154	65	20		10	25	14	14	W.4	2280- 2300	5331	95	20	
12	19	22	22	W.4	1602- 1603	1706	15	15	10	1	15	18	W.4	2823- 2840	9285	100	22		
7	23	27	21	W.4	1020- 1240	3264	40	21	3	10	2	16	13	W.4	2323- 2350	4733	105	19	
10	19	29	4	W.4	820- 870	458	0	3	4	6	20	16	19	W.4	3328- 3338	13613	105	12	
11	23	40	15	W.4	1000- 1120	6363	50	26	6	20	16	19	W.4	3328- 3338	13348	105	11		
6	19	40	16	W.4	1490- 1576	8142	105	53	5	10	9	17	13	W.4	2688- 2713	11448	160	20	
4	21	47	26	W.4	2512- 2660	5281	105	39	7	2	14	19	14	W.4	2754- 2774	12262	135	22	
14	5	49	22	W.4	1760- 1800	8253	90	35	5	10	25	19	14	W.4	2872- 2902	12898	95	10	
16	32	41	6	W.5	4498- 4503	1731	11	2	2	10	25	19	14	W.4	2872- 2902	15766	120	12	
8	6	42	5	W.5	4629- 4632	11855	80	9	6	10	27	19	14	W.4	2750- 2776	12105	135	28	
6	30	47	1	W.5	2934- 3014	3891	65	20		6	14	19	16	W.4	2806- 2831	10684	100	30	
6	9	48	2	W.5	2990- 3021	4692	100	44	10	27	19	16	W.4	2800- 2817	11649	160	41		
14	21	48	2	W.5	2908- 2925	6395	105	42	10	29	20	13	W.4	2572- 2590	17293	240	1		
10	27	48	2	W.5	2875- 2895	5965	100	39	6	10	8	20	14	W.4	2837- 2870	14489	205	46	
16	6	48	4	W.5	3102- 3128	5258	80	20	11	3	20	16	W.4	3032- 3040	14918	203	45		
4	30	48	4	W.5	3187- 3224	5848	90	32	11	32	20	16	W.4	2888- 3060	12999	105	4		
33	48	7	W.5	1000- 2280	24	0	0	4	6	14	21	2	W.4	2403- 2523	5663	60	17	13	
14	15	49	7	W.5	3298- 3305	4319	55	22	10	10	21	8	W.4	2520- 2590	5121	55	13		
9	35	50	9	W.5	3072- 3095	1537	0	3	5	4	20	21	9	W.4	2835- 2858	10386	75	5	
										10	25	21	10	W.4	2890- 2927	15035	125	2	
										11	27	21	13	W.4	2818- 2819	18833	145	15	
										7	13	22	8	W.4	2480- 2521	4782	60	15	
										7	15	22	8	W.4	2563- 2593	5073	50	16	
										7	15	22	8	W.4	2838- 2878	5962	45	9	12
										7	15	22	8	W.4	2612- 2617	4843	45	15	15
										11	8	22	11	W.4	2771- 2810	17325	190	15	
11	14	12	2	W.4	1516- 1536	7401	95	59	6	11	4	23	14	W.4	2605- 2611	12991	155	30	
4	18	65	23	W.5	3897- 3935	4511	110	30	12	11	4	23	14	W.4	2553- 2600	12454	170	46	
13	12	78	13	W.6	1400- 1435	907	0	5	2	10	6	24	12	W.4	2945- 2955	14488	120	18	11
10	9	75	8	W.6	2360- 2430	2098	35	7	4	6	15	25	5	W.4	2680- 2740	8439	115	28	
										10	4	25	11	W.4	2795- 2815	9185	85	30	
										16	23	25	12	W.4	3042- 3066	10543	190	21	
										10	12	25	23	W.4	4201- 4226	7157	150	45	10
										10	3	26	3	W.4	2419- 2427	8419	85	20	
										11	21	26	3	W.4	2449- 2457	9195	115	22	
										6	32	26	3	W.4	2468- 2478	9245	110	25	
										11	15	26	6	W.4	2629- 2640	11102	115	31	
										6	22	26	11	W.4	2813- 2891	8015	90	23	
										11	24	26	11	W.4	2740- 2770	7329	75	17	
										11	24	26	11	W.4	2740- 2770	9796	115	26	
										10	7	26	14	W.4	3030- 3088	12125	90	23	
										6	9	26	14	W.4	3014- 3069	12327	135	28	
										6	9	26	14	W.4	3274- 3291	12401	140	17	
										10	16	26	14	W.4	3236- 3256	12538	15		
										11	18	27	3	W.4	2500- 2513	9555	85	15	
										11	4	27	4	W.4	2462- 2485	10192	140	29	
										6	22	27	13	W.4	3100- 3130	8416	110	27	
										6	22	27	13	W.4	3110- 3140	8413	80	22	
										6	15	27	15	W.4	3525- 3555	13173	110	13	
										10	1	27	21	W.4	4150- 4180	5997	135	33	
										10	12	27	21	W.4	4045- 4075	6800	145	33	
										7	14	27	21	W.4	3920- 3950	6412	180	41	
										10	5	28	5	W.4	2615- 2680	11520	100	29	12
										10	27	28	7	W.4	2670- 2697	10570	125	30	
										6	34	28	9	W.4	2852- 2877	9596	130	35	
										6	31	29	4	W.4	2525- 2533	9016	175		
										10	14	29	20	W.4	3549- 3600	7730	180	49	11
										11	5	40	19	W.4	3873- 3933	19711	235	49	
										7	34	41	19	W.4	3500- 3550	22361	260	45	
										7	18	41	23	W.4	4401- 4419	21984	230	46	
										11	25	42	19	W.4	3158- 3200	12408	150	35	

Lower Cretaceous, Lower Colorado Group

(cont.)

Location	Interval	Cl	Br	I	B
7 31 44 17 W.4	2930- 2946	34221	280	28	
7 31 44 17 W.4	2887- 2915	36078	340	34	12
6 17 45 19 W.4	3135- 3170	42180	320	30	
7 12 45 20 W.4	3200- 3284	43552	330	28	
14 11 46 20 W.4	3231- 3234	39732	335	32	
9 15 46 20 W.4	3245- 3248	38592	300	32	
4 23 46 20 W.4	3251- 3255	36315	300	27	
10 2 47 20 W.4	3228- 3231	39050	360	29	
10 2 47 20 W.4	3228- 3231	39732	330	28	
16 9 47 20 W.4	3227- 3229	36531	315	34	
9 21 47 20 W.4	3210- 3216	37445	335	30	
6 33 47 20 W.4	3196- 3206	47180	300	27	
7 28 47 21 W.4	3351- 3365	31140	245	28	
7 28 47 21 W.4	3370- 3405	36112	275	33	
11 6 48 20 W.4	3215- 3218	32839	275	33	
12 7 48 20 W.4	3217- 3220	36201	305	30	
6 27 48 21 W.4	3252- 3260	34385	260	29	
3 33 48 21 W.4	3320- 3324	31390	280	32	
5 35 48 21 W.4	3265- 3270	35660	300	29	
3 20 49 21 W.4	3267- 3272	34110	320	33	

Lower Cretaceous, Lower Colorado Group

(cont.)

Location	Interval	Cl	Br	I	B
12 23 79 9 W.6	2622- 2636	8072	35	7	
11 24 79 9 W.6	2642- 2652	13548	105	20	
11 24 79 9 W.6	2576- 2586	13878	90	17	
13 21 79 11 W.6	2473- 2479	10506	230	19	18
10 9 80 9 W.6	2705- 2740	12573	65	13	
10 9 80 9 W.6	2780- 2897	10884	45	10	
10 30 80 10 W.6	2590- 2607	11683	85	18	
10 30 80 10 W.6	2676- 2680	11068	100	22	
6 32 80 10 W.6	2821- 2825	10625	80	15	
11 6 80 12 W.6	2381- 2391	3527	5	2	
11 6 80 12 W.6	2336- 2356	3707	10	15	
10 11 80 13 W.6	2355- 2372	2575	5	3	
10 11 80 13 W.6	2251- 2261	3740	10	4	
10 13 80 14 W.6	2232- 2246	2411	5	3	
10 13 80 14 W.6	2215- 2225	5471	45	10	
10 13 80 14 W.6	2270- 2285	5150	30	5	
10 13 80 14 W.6	2251- 2265	5402	45	11	

Lower Cretaceous, Mannville Group

Location	Interval	Cl	Br	I	B
14 5 49 22 W.4	3425- 3445	30978	290	45	
1 22 49 27 W.4	3900- 3930	23297	250	38	6
11 33 50 18 W.4	2544- 2550	42629	280	21	
7 11 50 22 W.4	3282- 3289	32535	270	30	
11 14 50 22 W.4	3292- 3293	40169	310	28	
2 23 50 22 W.4	3279- 3283	40838	295	25	
1 27 50 22 W.4	3294- 3299	36388	280	28	
7 30 50 22 W.4	3282- 3302	27180	275	37	7
13 16 50 24 W.4	3350- 3385	41347	350	43	7
6 4 51 22 W.4	3254- 3265	29690	255	28	
6 6 52 15 W.4	2125- 2170	36584	230	21	5
10 36 52 21 W.4	2823- 2840	45418	310	30	
3 32 52 26 W.4	3445- 3483	33905	310	39	
1 12 52 27 W.4	3590- 3605	32810	310	46	7
10 14 53 21 W.4	2776- 2786	47958	300	24	
10 15 53 21 W.4	2742- 2756	39659	270	23	
10 15 53 21 W.4	2742- 2756	44235	275	25	
10 15 53 21 W.4	2742- 2756	45752	265	25	
6 23 55 25 W.4	2850- 2870	42152	330	35	8
10 10 59 20 W.4	1875- 1938	31910	245	30	5
10 28 68 13 W.4	805- 825	14996	80	11	13
7 2 69 13 W.4	815- 850	16470	85	10	
10 22 74 25 W.4	1230- 1262	908	0	1	9
7 3 61 7 W.5	3100- 3185	21010	145	21	
12 28 63 16 W.5	5065- 5155	7406	70	12	12
4 15 65 24 W.5	5560- 5600	3103	30	7	
4 15 65 24 W.5	5525- 5560	4274	45	8	10
4 15 70 2 W.5	1225- 1480	15024	110	11	13
10 7 77 23 W.5	1291- 1344	10813	55	10	10
6 8 82 25 W.5	1404- 1420	12252	130	21	17
10 9 75 8 W.6	3790- 3825	14118	80	12	8
6 33 77 11 W.6	3141- 3171	9483	20	5	4
10 1 78 12 W.6	3329- 3342	12353	50	11	
10 1 78 12 W.6	3292- 3307	12705	55	10	
6 11 78 13 W.6	3147- 3273	8400	45	7	
6 11 78 13 W.6	3147- 3273	8084	95	15	
2 12 78 13 W.6	3130- 3158	10667	50	8	
13 12 78 13 W.6	3355- 3373	11406	85	12	
13 12 78 13 W.6	3380- 3390	11937	45	10	
13 12 78 13 W.6	3455- 3466	11353	45	8	
1 6 1 9 W.4	3008- 3029	1369	13	0	5
1 6 1 9 W.4	2694- 2704	117	0	0	
12 5 1 11 W.4	2474- 2496	4106	0	1	
6 18 1 13 W.4	2330- 2350	1370	0	1	
10 12 1 14 W.4	2272- 2300	855	0	0	
6 15 1 16 W.4	2520- 2570	1653	0	0	
12 3 1 17 W.4	2760- 2780	735	0	1	
12 9 1 17 W.4	2991- 3005	649	0	0	
7 16 1 20 W.4	3720- 3800	3129	0	1	
7 29 2 8 W.4	2830- 2850	685	4	0	7
11 21 2 9 W.4	2230- 2309	443	0	0	
11 9 3 8 W.4	2750- 2785	2786	5	1	5
10 18 4 5 W.4	3307- 3315	1016	0	0	
10 21 4 6 W.4	3016- 3040	3558	0	1	
2 23 4 17 W.4	3060- 3085	872	0	0	
4 13 6 3 W.4	4100- 4115	2916	0	3	11
6 9 7 8 W.4	3008- 3058	7395	50	7	
6 9 7 8 W.4	3016- 3064	1628	0	3	
10 14 7 8 W.4	3040- 3067	5286	0	3	
6 11 7 10 W.4	3002- 3025	2344	0	1	
8 18 7 16 W.4	3300- 3322	2731	0	1	
10 25 7 16 W.4	3217- 3241	1013	0	1	
4 18 9 17 W.4	3241- 3249	861	0	0	
16 14 10 16 W.4	3150- 3170	1028	0	1	5
16 32 12 16 W.4	3382- 3455	1981	0	1	
6 29 12 18 W.4	3568- 3612	3912	0	2	
11 32 12 18 W.4	3615- 3633	4571	5	3	15
4 30 14 13 W.4	3231- 3250	1566	0	1	
4 30 14 13 W.4	3234- 3310	1576	0	1	
10 28 15 11 W.4	3205- 3230	2234	0	1	
10 14 15 18 W.4	3674- 3705	11038	105	9	
4 26 16 12 W.4	3190- 3245	2946	0	2	
4 26 16 12 W.4	3165- 3180	3074	0	2	
12 26 16 12 W.4	3184- 3190	2937	0	2	
8 9 16 13 W.4	3365- 3404	3353	0	1	
8 9 16 13 W.4	3404- 3442	2935	0	1	
4 10 16 13 W.4	3252- 3300	3031	0	2	
2 11 16 21 W.4	4409- 4421	10022	30	3	
14 3 17 12 W.4	3280- 3372	3697	0	1	
10 4 17 12 W.4	3214- 3255	7018	0	2	

## Lower Cretaceous, Mannville Group

(cont.)

Location	Interval	Cl	Br	I	B
4 21 17 12 W.4	3213- 3266	3219	0	1	
10 26 17 13 W.4	3258- 3264	6771	30	1	
13 36 17 13 W.4	3245- 3246	13333	75	3	45
12 5 18 12 W.4	3333- 3345	6104	20	3	
6 13 18 13 W.4	3282- 3322	8389	60	1	
12 23 18 13 W.4	3259- 3261	8090	30	3	
10 30 18 13 W.4	3342- 3346	9466	30	4	24
12 34 19 9 W.4	3360- 3400	5784	0	2	
11 20 19 13 W.4	3293- 3319	7560	40	4	
11 20 19 13 W.4	3292- 3293	6925	40	2	27
8 1 19 14 W.4	3343- 3359	9348	55	4	
12 28 20 8 W.4	2841- 2866	3003	0	1	
14 32 20 8 W.4	3047- 3059	3971	13	2	
2 10 20 9 W.4	3207- 3213	8021	30	3	23
8 10 20 9 W.4	3290- 3322	8219	20	3	
14 10 20 9 W.4	3280- 3350	7495	60	1	
10 29 20 9 W.4	3345- 3400	7853	40	3	
2 11 20 12 W.4	3235- 3287	12704	160	7	
12 21 20 18 W.4	3916- 3966	17044	90	7	
12 21 20 18 W.4	3920- 3929	17107		7	
6 5 21 8 W.4	3226- 3281	4563	0	1	
8 6 21 8 W.4	3138- 3192	3088	50	3	
10 10 21 8 W.4	3162- 3195	3332	0	3	
11 17 21 9 W.4	3200- 3275	13752	125	11	
10 36 21 9 W.4	3206- 3297	7698	40	2	20
10 36 21 9 W.4	3169- 3202	5105	19	2	
10 25 21 10 W.4	3297- 3375	11657	150	7	
10 5 21 13 W.4	3370- 3417	9902	75	4	
10 15 21 14 W.4	3435- 3493	17469	155	11	
10 21 21 14 W.4	3438- 3453	17554	85	6	
11 1 22 8 W.4	3290- 3360	4252	10	3	
7 15 22 8 W.4	3191- 3202	4034	0	3	
12 32 22 8 W.4	2917- 2947	6147	55	7	
6 6 22 14 W.4	3320- 3356	14730	105	9	
6 1 22 15 W.4	3430- 3460	8938	50	3	
6 1 22 15 W.4	3430- 3460	10564	90	6	
6 12 22 15 W.4	3419- 3434	15671	60	5	
6 4 23 19 W.4	4275- 4295	12821	60	10	
2 15 24 14 W.4	3427- 3438	14074	75	5	
11 10 25 10 W.4	3135- 3167	4147	0	3	
11 21 26 3 W.4	2990- 3010	8256	65	17	
11 15 26 6 W.4	3190- 3265	9539	70	12	
10 13 26 10 W.4	3315- 3326	6710	20	6	
10 4 26 11 W.4	3335- 3352	5315	10	3	
6 22 26 11 W.4	3423- 3440	5970	10	3	
11 24 26 11 W.4	3340- 3350	7286	35	6	
11 24 26 11 W.4	3351- 3359	8094	35	4	
10 29 26 15 W.4	3856- 3878	969	75	13	
10 28 27 20 W.4	4430- 4447	7068	50	15	16
10 1 27 21 W.4	4604- 4610	9785	55	11	
11 12 27 21 W.4	4494- 4568	8275	50	12	
10 18 29 22 W.4	4566- 4586	15667	90	12	
10 10 29 23 W.4	4865- 4895	16788	35	4	
3 35 31 22 W.4	4866- 4868	12191	50	7	
10 27 33 4 W.4	2722- 2920	3343	35	19	8
6 24 39 19 W.4	4100- 4172	12918	55	3	
4 27 40 11 W.4	3042- 3072	34067	165	13	
16 29 40 11 W.4	3073- 3078	47385	210	10	34
1 32 40 11 W.4	3095- 3098	44054	180	5	
12 12 40 13 W.4	3261- 3343	35681	160	14	33
11 23 40 15 W.4	3140- 3160	54285	330	11	
16 27 40 15 W.4	3422- 3476	50680	385	10	
6 35 40 15 W.4	3331- 3356	41567	310	19	
6 35 40 15 W.4	3371- 3400	54529	320	16	
6 35 40 15 W.4	3470- 3495	47735	370	10	

## Lower Cretaceous, Mannville Group

(cont.)

Location	Interval	Cl	Br	I	B
7 7 40 16 W.4	3715- 3745	48608	310	11	
6 17 40 16 W.4	3635- 3678	65942	350	18	
6 17 40 16 W.4	3735- 3765	68801	480	8	
6 19 40 16 W.4	3695- 3725	57231	385	18	
6 19 40 16 W.4	3800- 3825	65153	485	9	
6 20 40 16 W.4	3740- 3750	53908	375	7	
6 20 40 16 W.4	3740- 3750	61176	430	7	
6 20 40 16 W.4	3630- 3657	60375	330	16	
6 20 40 16 W.4	3740- 3750	63165	430	10	
6 20 40 16 W.4	3630- 3657	54547	320	15	
10 13 40 17 W.4	3723- 3755	57125	330	16	
10 13 40 17 W.4	3840- 3855	65225	540	9	
7 34 40 17 W.4	3710- 3740	58758	435	15	
6 19 40 18 W.4	4050- 4070	39407	315	11	
6 19 40 18 W.4	4150- 4185	65327	520	9	
11 5 40 19 W.4	4343- 4373	41227	285	4	
10 3 40 23 W.4	4772- 4792	90027	545	14	
1 5 41 11 W.4	3090- 3116	43876	195	13	
16 28 41 12 W.4	2965- 2985	52720	280	13	20
4 28 41 13 W.4	3043- 3053	48859	315	8	
10 20 41 17 W.4	3770- 3870	64335	375	9	
11 19 41 18 W.4	3853- 3878	30210	195	8	
10 21 41 18 W.4	3730- 3785	30558	175	5	
7 10 41 25 W.4	5310- 5328	49724	230	6	
4 11 43 1 W.4	2003- 2007	51886	205	9	
10 22 43 11 W.4	2642- 2674	38605	300	14	
10 22 43 11 W.4	2766- 2803	44513	330	15	
7 2 43 19 W.4	3710- 3749	32203	195	12	
8 25 44 1 W.4	1819- 1828	79472	260	19	36
11 3 44 13 W.4	3018- 3034	43054	205	9	
10 20 44 14 W.4	3022- 3057	42794	350	21	
10 20 44 14 W.4	3105- 3120	45082	345	16	
11 3 44 17 W.4	3447- 3472	57874	380	10	
11 30 44 17 W.4	3379- 3389	27999	325	15	
11 30 44 17 W.4	3421- 3439	26778	305	14	
11 30 44 17 W.4	3522- 3547	58487	455	12	
11 30 44 17 W.4	3420- 3422	58918	310	7	
7 31 44 17 W.4	3358- 3368	41597	290	9	
7 31 44 17 W.4	3430- 3440	54526	360	7	
10 13 44 23 W.4	4510- 4550	74014	385	0	
2 25 45 6 W.4	2056- 2064	110776	515	24	76
7 32 45 6 W.4	2252- 2257	61513	370	16	55
1 34 45 6 W.4	2109- 2114	85539	470	24	41
1 34 45 6 W.4	2109- 2114	52860	280	8	
7 12 45 20 W.4	3803- 3855	62401	355	13	
10 30 46 20 W.4	3948- 3982	63164	395	13	
10 30 46 20 W.4	4005- 4020	63835	305	8	
6 14 46 22 W.4	4064- 4084	51877	260	7	
1 5 47 5 W.4	2209- 2215	137265	620	30	
8 5 47 5 W.4	2194- 2209	131573	730	38	86
11 5 47 5 W.4	2182- 2207	53180	265	21	
4 4 47 21 W.4	4050- 4092	36939	190	7	
7 28 47 21 W.4	4055- 4065	67756	360	17	
7 28 47 21 W.4	3973- 3985	53579	255	10	
7 28 47 21 W.4	3829- 3872	34313	180	7	
10 11 47 26 W.4	4790- 4826	28743	135	5	
10 10 48 6 W.4	1855- 1880	49737	320	13	
10 28 48 6 W.4	1829- 1840	46700	345	18	
13 30 48 22 W.4	4170- 4227	66999	300	12	
13 30 48 22 W.4	4100- 4115	64299	285	13	
1 33 48 26 W.4	4480- 4513	42269	205	6	
7 30 49 6 W.4	1897- 1923	50428	300	16	
3 31 49 6 W.4	2112- 2134	51047	250	10	
3 31 49 6 W.4	1810- 1820	22316	110	7	
11 36 49 17 W.4	2963- 2981	57404	260	15	

Lower Cretaceous, Mannville Group (cont.)						Lower Cretaceous, Mannville Group (cont.)					
Location	Interval	Cl	Br	I	B	Location	Interval	Cl	Br	I	B
11 36 49 17 W.4	2576- 2620	22442	145	8		11 5 69 13 W.4	1386- 1396	14043	40	15	12
10 9 49 18 W.4	3314- 3336	52486	345	2		16 22 82 4 W.4	747- 930	7435	0	0	5
11 26 50 1 W.4	1870- 1874	69604	385	2		6 13 91 18 W.4	549- 677	494	0	0	2
16 15 50 2 W.4	1984- 1988	53825	550	23		10 26 29 2 W.5	7980- 8038	34091	95	5	
6 4 50 14 W.4	2730- 2770	46473	410	29		6 17 37 3 W.5	7605- 7635	9365	20	4	
11 33 50 18 W.4	2920- 2950	49734	255	16		6 17 37 3 W.5	7525- 7617	15567	45	6	
7 32 50 20 W.4	4026- 4042	65140	340	15		4 35 38 4 W.5	7423- 7459	25759	110	21	
16 13 50 21 W.4	3560- 3576	58102	290	18		6 7 39 3 W.5	7110- 7148	19659	145	13	
16 13 50 21 W.4	3072- 3090	36280	305	16		4 32 39 3 W.5	7108- 7111	25660	110	11	
11 19 50 22 W.4	3868- 3902	54697	250	15		6 13 39 4 W.5	7117- 7145	23511	90	12	
11 19 50 22 W.4	3902- 3969	53892	405	12		2 21 48 1 W.5	5102- 5110	55522	210	11	
11 19 50 22 W.4	3962- 3984	54105	275	14		2 21 48 1 W.5	5102- 5110	55663	230	10	
11 19 50 22 W.4	3992- 4021	58120	250	14		2 21 48 1 W.5	5102- 5110	55011	220	10	
7 30 50 22 W.4	3812- 3838	51711	245	13		2 21 48 1 W.5	5102- 5110	54588	200	10	
7 30 50 22 W.4	4010- 4026	25656	115	6		2 21 48 1 W.5	5102- 5110	55240	205	9	
10 36 50 23 W.4	3800- 3838	51297	235	12		10 3 51 7 W.5	5755- 5819	27847	120	7	
10 36 50 23 W.4	3855- 3882	30851	260	13		10 3 51 7 W.5	5755- 5819	26755	110	6	
10 36 50 23 W.4	3947- 3972	61281	145	9		7 35 52 9 W.5	6216- 6420	35958	115	14	
13 16 50 24 W.4	3958- 3978	73124	315	15		7 35 52 9 W.5	6216- 6420	33089	110	15	
11 8 50 25 W.4	4080- 4130	55526	285	9		10 23 56 1 W.5	3885- 3892	60283	275	12	33
11 8 50 25 W.4	4190- 4235	62829	300	10		10 23 56 1 W.5	3885- 3892	55902	280	10	
10 31 51 24 W.4	3850- 3851	68929	340	20	65	10 23 56 1 W.5	6095- 6134	38440	95	13	
6 32 51 24 W.4	3899- 3932	64200	290	16	42	4 13 57 12 W.5	6095- 6134	40385	100	12	
10 10 51 25 W.4	4080- 4109	69657	235	11		4 13 57 12 W.5	6095- 6134	53751	190	12	
10 10 51 25 W.4	4080- 4118	68383	285	10		12 20 60 15 W.5	6142- 6165	28486	70	8	
16 28 51 26 W.4	4303- 4308	77280	385	15	58	9 22 63 15 W.5	5701- 5813	35866	95	11	
10 30 52 23 W.4	3696- 3707	65264	280	15	42	12 28 63 16 W.5	5945- 6040	35300	155	17	18
2 30 52 25 W.4	4021- 4051	62629	320	17	50	10 21 63 19 W.5	5831- 5963	36617	130	18	
10 19 52 26 W.4	4218- 4254	42417	285	10		2 2 63 22 W.5	7030- 7060	36224	110	17	
1 12 52 27 W.4	4335- 4345	65480	315	11		2 8 63 22 W.5	7131- 7161	32321	115	20	
14 17 53 23 W.4	3464- 3529	53099	270	15		4 13 65 24 W.5	6780- 6815	33272	140	23	
1 6 53 25 W.4	3960- 3964	67211	255	14		10 2 66 24 W.5	6698- 6748	31960	95	13	
2 29 53 25 W.4	3857- 3877	52608	360	14		4 7 68 7 W.5	3905- 3935	2010	0	9	
7 22 54 26 W.4	3775- 3815	41400	240	13		12 21 71 16 W.5	3505- 3680	25915	110	20	17
7 22 54 26 W.4	3990- 4051	63739	285	16		6 34 71 24 W.5	4597- 4618	21482	70	13	
16 17 55 20 W.4	3347- 3348	55296	185	15	37	3 26 72 24 W.5	4438- 4487	4510- 4590	13649	40	11
10 19 55 24 W.4	3569- 3579	51961	220	12		10 35 72 24 W.5	4481- 4521	21736	60	5	
6 23 55 25 W.4	3615- 3641	54790	240	14		10 35 72 24 W.5	4430- 4470	14666	40	10	
10 31 56 25 W.4	3432- 3444	35308	285	17		10 3 73 22 W.5	3816- 3835	13705	25	8	
10 31 56 25 W.4	3551- 3606	48046	240	16	31	10 3 73 22 W.5	3931- 4050	21607	55	8	
6 16 56 27 W.4	3810- 3830	56884	265	12	29	10 3 73 22 W.5	3931- 4050	9837	15	5	
11 21 57 5 W.4	1350- 1355	33423	170	16	17	2 27 74 23 W.5	3473- 3539	11640	10	5	
10 34 57 5 W.4	1397- 1412	33572	130	8		10 7 77 23 W.5	2550- 2598	19662	85	16	
4 3 57 20 W.4	2595- 2607	48338	210	10		10 11 77 23 W.5	2802- 2814	982	0	0	
4 3 57 20 W.4	2592- 2630	42387	255	12		12 15 78 11 W.5	1382- 1397	3527	0	1	
14 30 57 20 W.4	2811- 2825	26745	120	3		10 36 83 23 W.5	2140- 2190	16074	90	28	18
11 7 58 5 W.4	1634- 1800	27074	105	9		9 28 87 20 W.5	860- 890	18505	130	20	
7 18 58 19 W.4	2521- 2586	36866	125	9		10 23 71 1 W.6	4449- 4547	1013	0	2	
10 10 59 20 W.4	2460- 2520	34658	215	13		11 28 77 11 W.6	4938- 4942	851	0	0	
10 13 61 6 W.4	1024- 1036	24113	85	9		12 23 79 9 W.6	4401- 4405	931	0	0	
7 9 62 19 W.4	2270- 2300	9804	30	2		11 24 79 9 W.6	4381- 4385	370	0	0	
10 9 62 21 W.4	2265- 2300	31777	165	9		10 30 80 10 W.6	4422- 4426	1216	0	0	
12 12 63 3 W.4	904- 908	18504	70	10	9	10 30 80 10 W.6	4376- 4380	976	0	0	3
10 26 63 21 W.4	2397- 2465	29997	120	10		10 30 80 10 W.6	4029- 4033	9762	25	15	
10 26 63 21 W.4	2468- 2498	32115	125	11		6 32 80 10 W.6	4629- 4633	999	0	1	6
10 26 63 21 W.4	2573- 2600	30651	135	11		6 32 80 10 W.6	4235- 4239	10182	35	16	5
9 22 64 3 W.4	1128- 1140	18330	60	8		10 13 81 13 W.6	3639- 3694	10690	30	20	
10 8 66 19 W.4	1459- 1501	16320	45	8	12	10 13 81 13 W.6	2531- 2588	494	0	1	3
6 25 66 23 W.4	1632- 1661	15793	75	7		10 13 81 13 W.6	4155- 4187	533	0	1	4
6 25 66 23 W.4	1946- 1957	21983	105	12		2 11 82 2 W.6	2824- 2940	66311	175	18	53
6 8 66 25 W.4	2610- 2730	28798	100	12		10 29 96 3 W.6	2053- 2108	7992	35	12	6
11 26 67 23 W.4	1415- 1460	17389	130	13							
6 6 67 24 W.4	1972- 1977	25793	95	8							
7 2 69 13 W.4	1475- 1563	12701	35	6							
7 2 69 13 W.4	965- 980	14518	70	10							

Jurassic						Carboniferous, Rundle Group (cont.)					
Location	Interval	Cl	Br	I	B	Location	Interval	Cl	Br	I	B
4 7 1 5 W.4	3500- 3534	1324	0			12 24 1 10 W.4	3202- 3230	333	0	0	
10 21 4 6 W.4	3260- 3300	530	0	0		12 5 1 11 W.4	2800- 2825	110	0	7	
2 34 5 16 W.4	3186- 3196	3983	5	3		2 23 1 12 W.4	2806- 2866	167	0	0	
4 13 6 3 W.4	4122- 4300	2441	50		42	6 15 1 16 W.4	2760- 2800	212	0	0	
4 13 6 3 W.4	4303- 4370	695	0	1		10 3 1 17 W.4	2920- 2934	1259	0	1	
11 32 6 9 W.4	3140- 3160	2702	5	3		12 3 1 17 W.4	2897- 2910	757	0	0	
10 32 7 15 W.4	3259- 3286	814	0	0		13 3 1 17 W.4	3101- 3108	773	0	0	
10 16 7 16 W.4	3320- 3330	725	0	0		11 5 1 17 W.4	3091- 3100	791	0	1	
8 6 9 16 W.4	3178- 3190	1120	0	0		11 1 1 18 W.4	3295- 3332	904	0	0	4
10 3 9 19 W.4	3644- 3700	1879	0	1	13	11 21 2 9 W.4	2830- 2880	886	0	0	
11 14 10 16 W.4	3202- 3223	1144	7	1	5	10 13 3 10 W.4	2934- 3000	1394	0	7	
7 10 11 13 W.4	3035- 3037	1019	0			10 18 6 16 W.4	3270- 3301	732	0	1	
7 10 11 13 W.4	3035- 3037	985	0	4		10 27 6 16 W.4	3300- 3320	853	0	0	
4 13 11 14 W.4	2975- 3007	1843	0	1		10 31 7 15 W.4	3280- 3300	763	0	0	
1 8 13 13 W.4	3130- 3139	1428	6	1	4	10 16 7 16 W.4	3366- 3376	757	0	0	
6 7 39 3 W.5	7179- 7190	21741	105	12		2 3 9 17 W.4	3360- 3390	2590	0	1	
6 7 39 3 W.5	7216- 7240	14589	65	12		8 26 9 17 W.4	3233- 3358	3074	5	1	
12 26 56 13 W.5	6188- 6210	36503	190	25		4 13 11 14 W.4	3011- 3031	1042	0	0	
2 18 57 13 W.5	6392- 6427	41446	195	21		11 35 13 17 W.4	3342- 3350	12190	58	5	14
10 22 59 13 W.5	5708- 5770	16583	65	6		7 17 16 18 W.4	3905- 3925	19472	170	7	
						6 10 16 26 W.4	6843- 6920	15161	80	3	
						6 10 16 26 W.4	6896- 6921	15409	80	4	
						4 21 17 12 W.4	3300- 3342	3385	5	1	
						6 34 19 9 W.4	3332- 3362	7107	25	2	3
						3 8 19 11 W.4	3302- 3311	8783	40	4	
						7 8 19 11 W.4	3345- 3346	10201	50	7	30
						14 14 19 14 W.4	3394- 3464	12452	70	2	
						10 14 19 14 W.4	3415- 3437	9501	50	3	
						10 10 20 9 W.4	3304- 3324	8798	35	4	
						10 16 21 15 W.4	3527- 3567	10968	170	8	
						6 15 21 16 W.4	3675- 3713	8661	35	5	
						16 32 31 24 W.4	5220- 5358	33093	130	20	28
						11 27 30 3 W.5	8340- 8500	42364	195	16	
						12 32 33 5 W.5	9470- 9500	31868	220	13	
						12 32 33 5 W.5	9475- 9495	35962	420		
						12 32 33 5 W.5	9475- 9495	36094	220	19	
						3 9 34 5 W.5	9198- 9208	39306	275	18	
						6 16 35 5 W.5	9070- 9083	45245	200	19	
						4 2 35 6 W.5	9633- 9643	26675	265	16	
						4 2 35 6 W.5	9633- 9643	32490	295	12	
						4 2 35 6 W.5	9643- 9665	27534	330		
						10 27 35 6 W.5	9440- 9460	42429	170	15	
						4 27 36 3 W.5	7580- 7795	7668	30	5	
						8 15 37 3 W.5	7435- 7467	17813	60	13	
						12 20 37 3 W.5	7555- 7580	27268	110	15	
						10 29 37 7 W.5	9515- 9556	39530	220	11	
						10 20 38 5 W.5	8135- 8175	35294	250	18	
						4 4 40 3 W.5	7105- 7113	22750	90	14	
						8 5 40 3 W.5	7091- 7095	34260	95	15	
						6 30 44 4 W.5	7126- 7185	31214	120	12	
						11 4 45 5 W.5	7153- 7205	16166	60	7	
						6 22 45 5 W.5	6976- 7015	62919	260	13	
						11 10 49 16 W.5	10180-10240	28078	100	2	152
						10 18 50 17 W.5	10022-10152	34108	225	10	
						10 21 51 9 W.5	6500- 6570	62849	265	18	
						10 21 51 9 W.5	6495- 6521	61223	250	17	
						4 29 55 19 W.5	9470- 9494	60383	180	15	
						10 25 56 9 W.5	5116- 5121	60035	230	13	
						4 29 60 19 W.5	7720- 7800	74409	200	11	
						4 2 63 10 W.5	4872- 4920	59672	60	10	



Carboniferous, Rundle Group (cont.)							Upper Devonian, Winterburn Group								
Location		Interval	Cl	Br	I	B	Location		Interval	Cl	Br	I	B		
10 25	63 10	W.5	5081- 5136	26754	80	8	10 4	19 11	W.4	4132- 4162	3275	5	1	18	
6 5	63 22	W.5	7751- 7811	84003	180	19	10 21	19 17	W.4	4860- 4880	7773	10	2	23	
2 16	64 11	W.5	5462- 5559	38543	125	16	64	10 14	29 20	W.4	5300- 5320	36161	320	6	
3 9	70 13	W.5	3836- 3856	44925	170	14	10 25	29 20	W.4	5419- 5424	28100	210	5	71	
10 21	72 13	W.5	3137- 3289	19699	65	9	4 35	29 20	W.4	5423- 5440	28572	210	2		
4 31	73 13	W.5	2699- 2756	27287	95	16	10 13	29 21	W.4	5530- 5640	41520	340	7		
4 31	73 13	W.5	2699- 2756	18313	170	32	40	10 13	29 21	W.4	5559- 5575	61401	500	8	117
9 7	78 15	W.5	2600- 2616	22894	35	17	20	2 12	30 20	W.4	5379- 5394	40400	335	5	
10 27	81 9	W.5	1690- 1711	4309	15	8	4 17	30 20	W.4	5670- 5745	62095	455	7		
13 8	81 19	W.5	2792- 2827	39800	150	20	10 11	33 17	W.4	4990- 5015	113090	200	15		
10 15	101 8	W.6	2545- 2590	10991	55	12	10	10 11	33 17	W.4	4982- 5008	100055	540	9	
Carboniferous, Banff Formation							14 22	33 23	W.4	6200- 6260	139137	1040	18		
Location		Interval	Cl	Br	I	B	15 27	34 20	W.4	5459- 5509	81642	730	15		
10 4	19 11	W.4	3752- 3800	11593	20	3	5 26	34 21	W.4	5680- 5715	104363	935	13		
10 10	20 9	W.4	3600- 3680	12442	50	3	2 1	35 20	W.4	5530- 5551	82869	705	13		
11 31	23 9	W.4	3399- 3440	3292	5	3	14	12 25	35 22	W.4	6037- 6087	140957	1245	24	
6 15	25 5	W.4	3390- 3440	5430	40	3	10 14	35 25	W.4	6985- 7065	141315	1010	23		
11 21	26 3	W.4	3070- 3110	7421	55	6	6 26	36 20	W.4	5160- 5161	69116	615	12		
10 20	26 5	W.4	3220- 3255	9714	75	7	4 8	37 20	W.4	5400- 5423	40223	360	6		
11 15	26 6	W.4	3310- 3357	9816	65	8	4 8	37 20	W.4	5400- 5423	62353	530	9		
7 9	27 4	W.4	3190- 3210	10187	85	10	4 8	37 20	W.4	5400- 5423	70342	580	12		
10 19	29 4	W.4	2925- 2970	10303	80	17	13	2 18	37 20	W.4	5407- 5420	75315	910	19	
6 31	29 4	W.4	3028- 3052	5753	90	16	13 17	37 23	W.4	5890- 5894	103443	775	17		
11 35	55 4	W.5	4281- 4291	47174	350	10	6 3	38 20	W.4	5229- 5276	121115	830	17		
12 20	56 5	W.5	4351- 4390	40978	150	9	14 29	39 23	W.4	5945- 5980	132229	980	13		
10 14	59 7	W.5	4230- 4277	27282	110	11	14 29	39 23	W.4	6025- 6045	130146	975	19		
2 21	60 7	W.5	4415- 4445	40757	170	21	1 15	39 26	W.4	6842- 6897	91908	580	9		
2 14	82 8	W.5	1742- 1872	1624	0	1	1	6 35	40 15	W.4	3700- 3791	63648	450	13	86
Upper Devonian, Wabamun Group							9 10	40 24	W.4	6205- 6207	125527	1000	12		
Location		Interval	Cl	Br	I	B	11 22	41 18	W.4	4360- 4560	69425	480	12		
11 22	41 18	W.4	3840- 3875	24884	170	9	2 10	44 22	W.4	4967- 4976	99808	535	13		
4 28	42 22	W.4	4790- 4835	40797	250	6	31	4 14	44 22	W.4	5108- 5117	95921	535	10	
5 29	47 14	W.4	2928- 3012	35983	215	10	10 23	44 28	W.4	6890- 6975	108159	550	16		
1 12	52 27	W.4	4596- 4612	83614	460	11	7 1	45 22	W.4	4662- 4667	83224	490	16	101	
10 10	59 20	W.4	2596- 2613	34818	180	11	5 29	47 14	W.4	3168- 3200	35405	180	11	22	
7 34	62 20	W.4	2325- 2350	27790	135	12	18	1 33	48 26	W.4	5465- 5482	119066	515	14	
6 6	67 24	W.4	2625- 2650	19868	90	6	10 34	49 25	W.4	4955- 5025	76513	210	11		
7 34	31 1	W.5	8549- 8572	78490	170	3	315	10 22	49 26	W.4	5301- 5323	108018	645	16	
2 21	60 7	W.5	4928- 5063	96333	320	15	15 22	49 26	W.4	5305- 5318	106842	520	16		
2 21	60 7	W.5	5428- 5505	101027	320	20	110	6 33	49 26	W.4	5190- 5224	155269	1180	22	
10 2	66 24	W.5	9670- 9710	166358	425	20	140	3 2	49 27	W.4	5526- 5546	106530	595	23	
4 32	68 8	W.5	5440- 5488	93392	220	10	11 13	49 27	W.4	5413- 5440	107083	465	16		
4 21	69 22	W.5	7900- 7942	188952	350	20	4 23	49 27	W.4	5428- 5445	107734	425	12		
4 15	70 2	W.5	3004- 3038	39240	185	20	7 30	50 22	W.4	4375- 4435	57348	375	15	75	
4 15	70 2	W.5	2302- 2350	20212	160	23	7 30	50 22	W.4	4433- 4445	50386	325	10		
4 15	70 2	W.5	2287- 2296	33616	150	34	33	10 36	50 23	W.4	4400- 4460	61281	355	12	
4 15	70 2	W.5	2232- 2242	29858	110	14	13 16	50 24	W.4	4650- 4670	65098	405	12		
10 35	73 4	W.5	2377- 2511	6918	220	5	2 12	50 26	W.4	5038- 5084	85430	435	14		
10 31	76 6	W.5	3136- 3235	1195	20	1	63	1 16	50 26	W.4	5059- 5394	106733	695	17	
13 8	81 19	W.5	5821- 5871	148694	440	12	12 28	50 27	W.4	5250- 5280	85389	460	12		
11 31	82 1	W.5	1412- 1425	5709	15	3	5 9	51 26	W.4	4988- 5002	113654	595	24		
6 13	82 3	W.5	1565- 1625	7950	20	1	15 34	51 27	W.4	5105- 5107	131922	955	24		
12 29	86 7	W.5	2450- 2540	3388	0	1	5 18	52 21	W.4	4030- 4065	62337	310	16		
6 28	85 3	W.6	6210- 6410	122856	410	20	31	5 18	52 21	W.4	4065- 4075	56747	280	12	
							7 26	52 26	W.4	4640- 4648	69430	390	8		
							1 12	52 27	W.4	5100- 5150	62904	400	11		
							14 18	54 25	W.4	4632- 4688	62036	400	12		
							4 19	54 25	W.4	4437- 4463	66040	400	14		
							10 31	56 25	W.4	4197- 4252	73704	420	14		
							15 29	58 24	W.4	3682- 3710	57615	375	16		

Upper Devonian, Winterburn Group (cont.)										Upper Devonian, Woodbend Group (cont.)						
Location		Interval	Cl	Br	I	B	Location		Interval	Cl	Br	I	B			
6	13	60 17 W.4	2049- 2090	25474	95	10	1	17	56 21 W.4	3177- 3195	61682	210	12			
10	26	63 21 W.4	2670- 2700	27694	105	12 21	7	18	56 22 W.4	3644- 3660	58731	240	13			
12	28	67 18 W.4	1838- 1910	14196	350	11 26	4	5	56 25 W.4	4495- 4515	70564	560	17			
9	23	43 2 W.5	7407- 7465	120190	660	30 167	2	6	57 20 W.4	3124- 3130	63457	220	24			
16	16	50 2 W.5	5985- 6025	100251	435	19 144	14	30	57 20 W.4	3921- 3950	137204	390	20 63			
4	20	50 2 W.5	6085- 6135	118419	495	20	4	31	57 21 W.4	3186- 3187	57617	220	14			
2	21	60 7 W.5	5628- 5672	94921	425	27 144	3	4	58 21 W.4	3310- 3318	62559	275	15			
10	26	62 17 W.5	8300- 8358	68688	160	12	10	2	58 22 W.4	3196- 3254	60256	300	20			
10	18	63 19 W.5	8605- 8695	76923	210	13	5	20	58 22 W.4	3198- 3211	63343	250	13			
6	5	63 22 W.5	9890- 9950	108929	280	12	3	8	58 24 W.4	3930- 3960	41573	265	9			
4	15	70 2 W.5	3195- 3220	41559	245	15	3	8	58 24 W.4	3930- 3960	61065	380	18 79			
4	15	70 2 W.5	3245- 3275	65920	255	12 53	10	26	63 21 W.4	3250- 3310	41220	200	12 25			
6	34	71 24 W.5	8385- 8440	157082	425	25 148	10	26	63 21 W.4	3814- 4000	90870	240	16 44			
10	36	76 8 W.5	3830- 3910	44939	165	13	6	6	67 24 W.4	3201- 3280	36961	185	9 26			
							11	5	69 13 W.4	1473- 1476	10718	50	10 14			
							11	5	69 13 W.4	1480- 1520	9180	10	4 8			
							16	12	33 2 W.5	9170- 9201	150043	1015	19 296			
							2	1	34 2 W.5	9140- 9165	164674	1015	21			
							6	17	37 3 W.5	9445- 9464	130499	930	19			
							6	17	37 3 W.5	9390- 9450	133015	920	11			
							6	17	37 3 W.5	9464- 9484	127982	980	12 207			
							11	7	43 1 W.5	7852- 7854	149737	1120	23 235			
							6	32	56 20 W.5	12065-12085	133865	295	11 269			
							4	26	57 19 W.5	10840-11024	113248	280	20 205			
							10	22	60 17 W.5	9388- 9416	129413	480	22 360			
							10	22	60 17 W.5	9388- 9416	131885	480	17			
							10	22	60 17 W.5	9388- 9416	127604	470	15			
							11	6	63 25 W.5	11830-11831	154152	430	13 114			
							7	27	67 22 W.5	8708- 8712	171835	455	19 118			
							10	27	67 22 W.5	8707- 8742	149752	400	13			
							13	24	68 22 W.5	8491- 8496	171849	385	15			
							7	33	68 22 W.5	8608- 8622	154299	370	21 168			
							11	26	69 23 W.5	8766- 8771	145540	385	19 158			
							4	15	70 2 W.5	3687- 3715	40815	350	23 47			
							6	34	71 24 W.5	8451- 8482	151415	370	15			
							6	34	71 24 W.5	8628- 8675	156585	385	16			
							6	34	71 24 W.5	9052- 9100	163094	400	19			
							10	35	73 4 W.5	3342- 3384	43577	330	23 64			
							11	9	79 22 W.5	6718- 6719	157987	516	12 47			
							13	8	81 19 W.5	6112- 6140	119273	390	16 27			
							13	8	81 19 W.5	6155- 6190	151250	370	16			
							6	24	87 1 W.6	6490- 6538	119093	355	15			
							2	16	87 2 W.6	6344- 6368	119537	395	15 59			
Upper Devonian, Beaverhill Lake Formation																
Location		Interval	Cl	Br	I	B	Location		Interval	Cl	Br	I	B			
4	17	30 20 W.4	6960- 7070	195955	2115	27 305	15	16	44 13 W.4	4272- 4316	124178	1590	25 102			
6	7	57 9 W.4	2921- 2922	49144	80	5 17	14	30	57 20 W.4	4489- 4509	63705	225	3 17			
10	7	64 19 W.4	3679- 3784	90725	440	38 73	10	7	64 19 W.4	3679- 3784	90725	440	38 73			
6	22	64 19 W.4	3757- 3790	71285	345		6	22	64 19 W.4	3757- 3790	71285	345				
6	36	58 19 W.5	11280-11305	134694	340	14 206	12	34	62 12 W.5	8578- 8644	111028	300	39 109			
4	2	63 10 W.5	8119- 8205	128591	365	41	4	2	63 10 W.5	8119- 8205	128591	365	41			
2	31	63 12 W.5	9404- 9664	104317	260	26	2	31	63 12 W.5	9404- 9664	104317	260	26			
10	6	63 21 W.5	10890-10903	98375	155	24 920	2	2	63 22 W.5	11067-11077	164132	375	17 512			
10	10	66 11 W.5	8660- 8686	109396	250	27	10	10	66 11 W.5	8660- 8686	109396	250	27			
14	29	69 19 W.5	8904- 8950	131301	260	22 144	4	3	70 18 W.5	8909- 8945	127086	265	17			
4	14	83 10 W.5	5412- 5484	76838	305	7 18	4	14	83 10 W.5	5412- 5484	76838	305	7 18			
12	17	93 21 W.5	5190- 5240	100681	220	10 20	12	17	93 21 W.5	5190- 5240	100681	220	10 20			

Middle Devonian, Elk Point Group

Lower Paleozoic

Location					Interval	Cl	Br	I	B
10	31	54	12	W.5	10006-10026	87975	460	12	143
10	22	60	17	W.5	10062-10094	206361	575	17	149
10	22	60	17	W.5	10062-10094	183312	740	22	210
4	8	71	4	W.5	6115- 6139	178088	720	3	38
4	8	71	4	W.5	6115- 6139	175047	700	3	38
4	8	71	4	W.5	6115- 6139	171581	720	3	28
10	11	71	12	W.5	7625- 7685	142474	465	7	71
2	9	73	4	W.5	5308- 5318	170530	670	3	30
2	6	73	5	W.5	5492- 5614	156156	530	5	37
2	10	73	18	W.5	8106- 8116	174213	580	8	79
4	10	75	6	W.5	5740- 5865	136680	560	4	35
10	31	76	6	W.5	6272- 6288	200334	1200	12	46
2	16	80	9	W.5	5649- 5660	105132	375	9	33
2	33	87	5	W.5	5012- 5075	116528	480	6	8
10	6	91	1	W.5	4020- 4080	182235	100	2	22
12	17	93	21	W.5	5402- 5447	116833	415	18	40
6	15	98	11	W.5	5326- 5354	116306	330	7	40
10	22	100	17	W.5	4806- 4916	204371	395		44
6	32	114	23	W.5	3770- 3805	51088	260	13	34
1	18	119	21	W.5	4040- 4150	59468	270	7	100
10	21	120	23	W.5	3720- 3750	87376	400	15	100
10	21	120	23	W.5	4316- 4345	72329	350	15	172
6	33	113	7	W.6	5425- 5513	87613	320	12	132
3	27	117	4	W.6	4802- 4812	73198	335	8	117
2	7	117	10	W.6	5380- 5407	56371	300	4	112
10	11	118	4	W.6	4936- 4941	89725	395	12	144

Location					Interval	Cl	Br	I	B
4	13	6	3	W.4	8190- 8210	27246	110	1	24
6	34	71	24	W.5	10033-10061	184801	455	8	29

Granite Wash

Location					Interval	Cl	Br	I	B
4	15	70	2	W.5	6422- 6464	205952	125	14	21
10	23	81	9	W.5	5668- 5734	147996	680	11	41
12	26	81	9	W.5	5675- 5736	147402	700	12	25
12	27	81	9	W.5	5668- 5674	157469	720	13	41
4	34	81	9	W.5	5713- 5737	161662	690	11	38
2	6	88	21	W.5	6287- 6315	122955	510	17	38
6	15	98	11	W.5	5541- 5574	126052	400	14	35
6	15	98	11	W.5	5541- 5574	123227	400	14	55